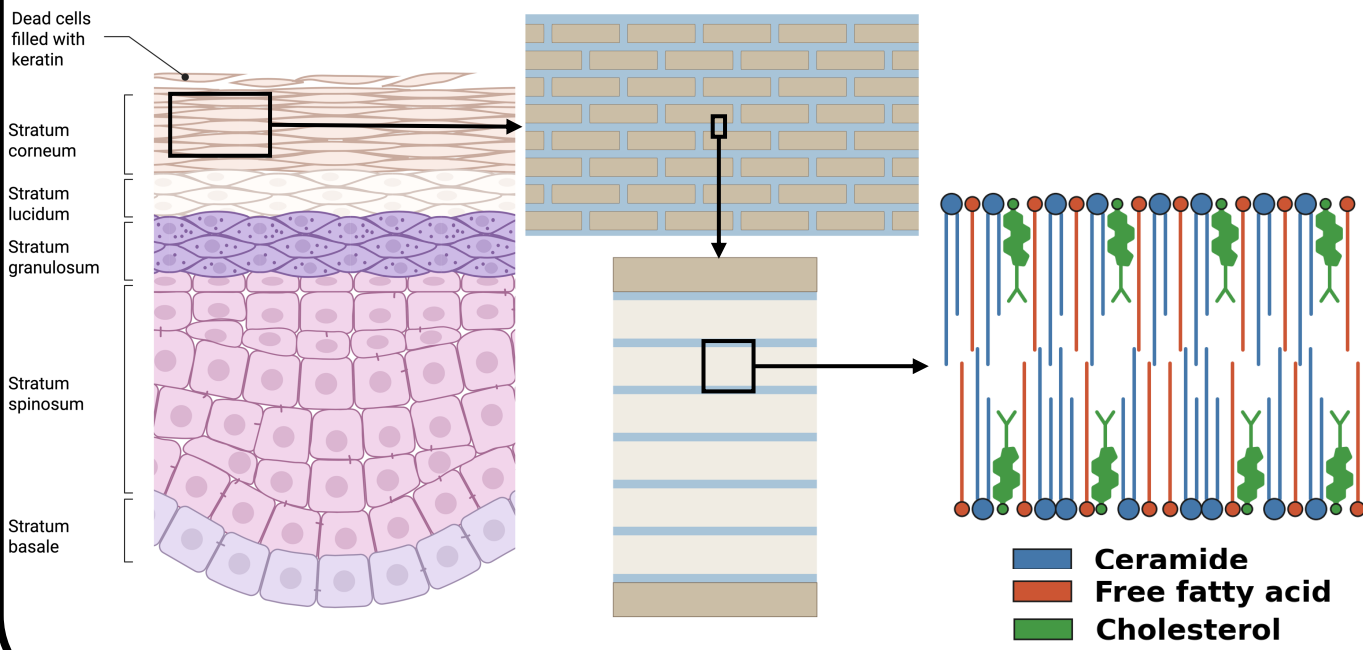


Molecular Dynamics of Synergistic Permeation Enhancement: Propylene Glycol and Isopropanol at the Skin Barrier

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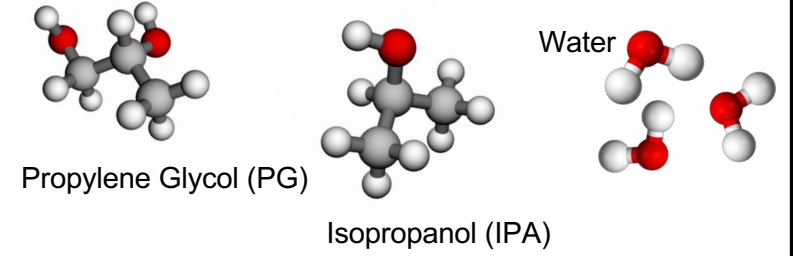
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Background: Stratum Corneum



- The skin's outermost layer, the stratum corneum (SC), is a crucial barrier against pathogens, water loss, and environmental damage, and is the main obstacle for topical drug delivery.
- The SC is formed by dense lipid lamellae comprised of ceramides, cholesterol, free fatty acids that surround the skin cells (corneocytes).

Permeation Enhancers

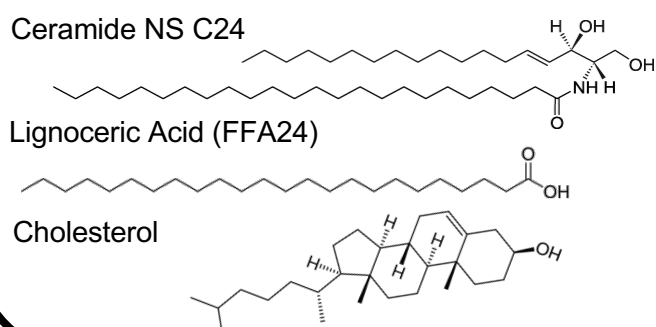


- Permeation enhancers (PEs) such as isopropanol (IPA) and propylene glycol (PG) are widely used to boost drug delivery through skin by disrupting the lipid barrier.¹
- How PEs trigger changes in SC lipid organisation and structure remains unclear and depends on chemistry.

Simulated Systems

Lipids in 1:1:0.5 molar ratio

- Ceramide NS C24 (CER NS)
- C24 Fatty Acid (FFA24)
- Cholesterol (CHOL)

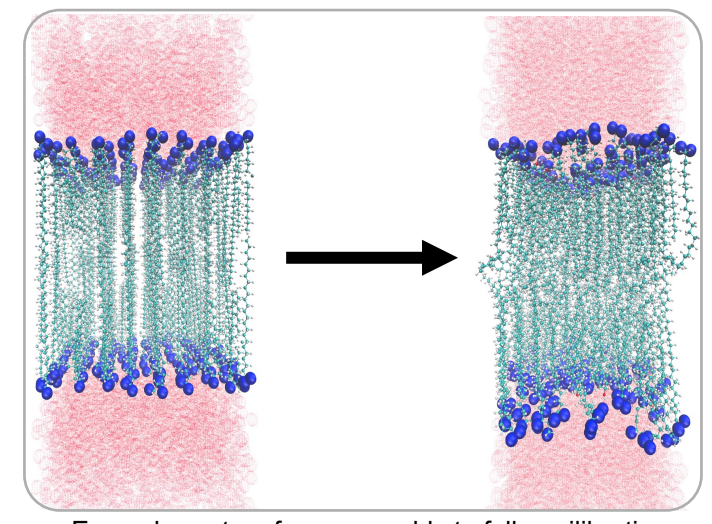


Solvent systems:

- Water + Isopropanol**
Concentrations 0, 10, 20 %_{w/w}
- Water + PG**
Concentrations 0, 20, 50, 70, 80, 90, 100 %_{w/w}
- Water + IPA/PG Mixtures**
- Total 19 simulated systems**
3 replicas of each system

Methods

- CHARMM based forcefields used for skin lipids, propylene glycol,² and isopropanol and *mTIP3P* used for water
- 64 lipids per leaflet assembled using MoSDeF³ and then solvated with water
- 50 ns of RWMD equilibration at 305-355 K⁴
- Water replaced as with PG/IPA/water mixtures and system re-equilibrated
- Production runs for 1 μ s at 305K and 1 atm
- Final 400 ns used for analysis

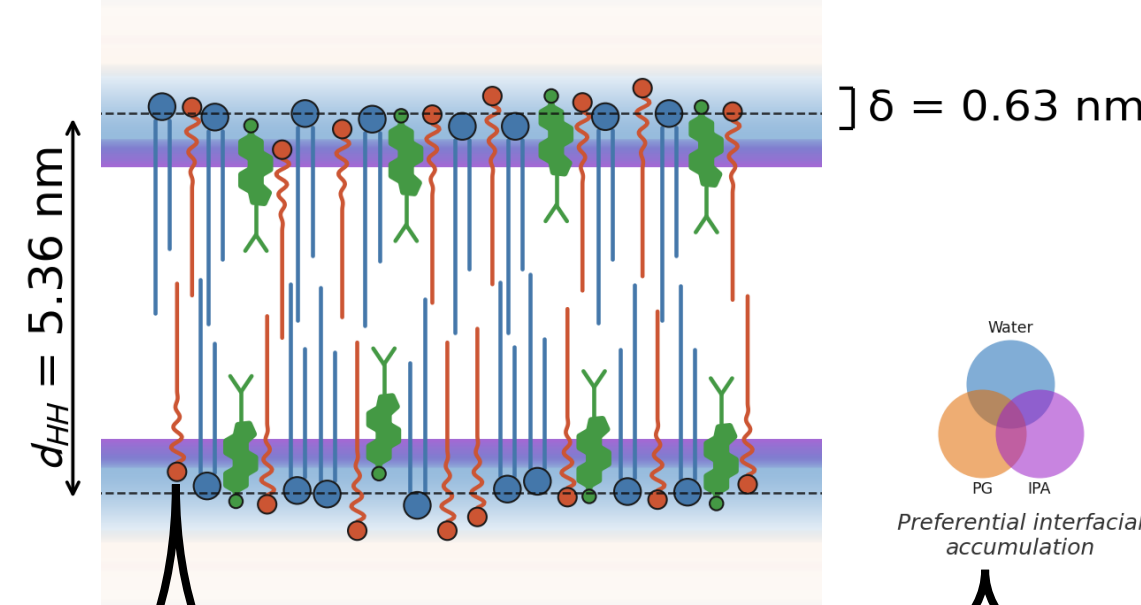
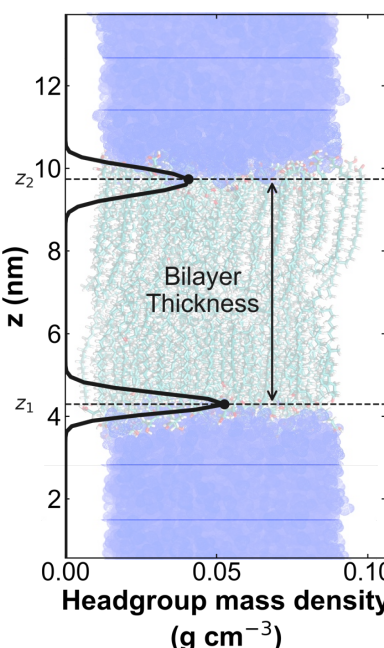


Example system from assembly to full equilibration
Headgroups - Blue, Lipid Tails - Teal, Solvent - Red

Results: Water 30 %_{w/w}, PG 50 %_{w/w}, IPA 20 %_{w/w}

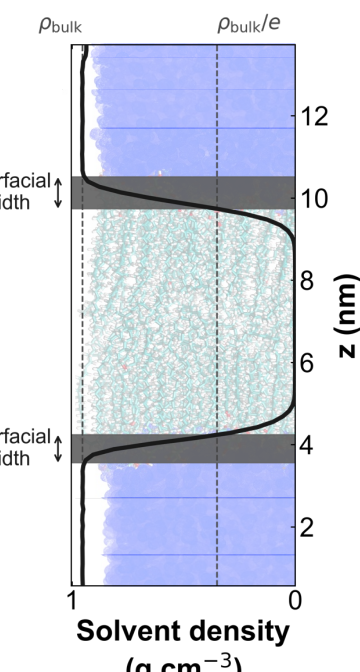
Bilayer Thickness (d_{HH})

d_{HH} and the complementary area per lipid (APL) track how solvent molecules perturb the head-group region. Reduced height with increased APL signals intercalation of solvent molecules in the headgroup region.

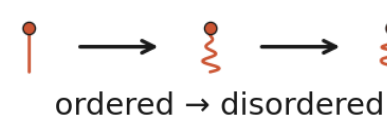


Interfacial Width (δ)

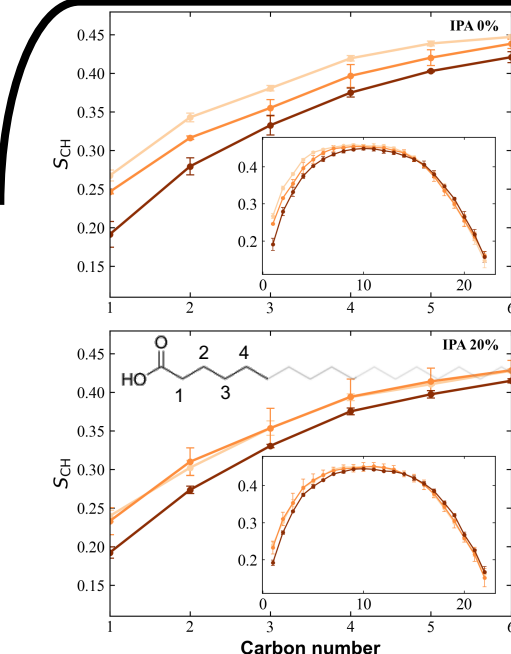
The interfacial width δ measures the sharpness of the lipid-solvent boundary, defined as the distance over which the solvent density rises from bulk/e to its bulk value. It is sensitive to how solvent molecules organise at the bilayer surface.



Tail Order Parameter (S_{CH})

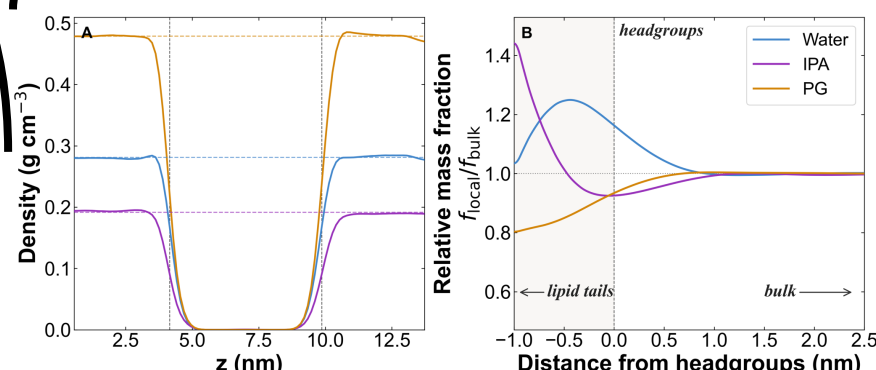


S_{CH} tracks the conformational order of the lipid chains, with the FFA24 tail being the most responsive to the PEs. The composition dependent drop near the head-group region signals disorder as the solvent perturbs the interface. The deeper regions remain largely unchanged (gel-phase).



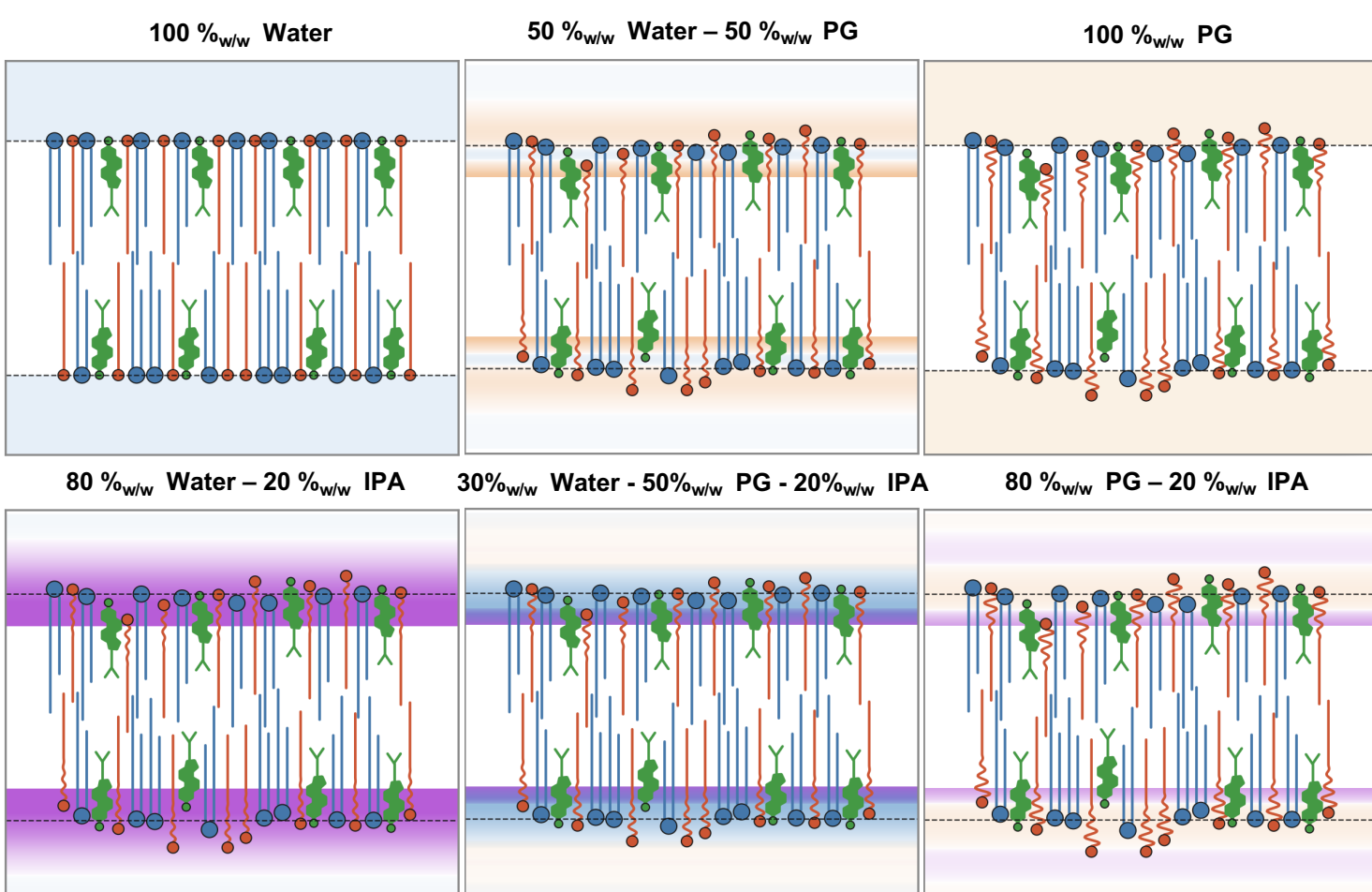
S_{CH} of FFA24 under different Solvent conditions: Deeper orange shading indicates increasing PG concentration (0%, 50%, 80%(bottom)/100%(top))

Interfacial Accumulation



Density profiles show where each solvent species accumulates. (A) Mass density of water, PG, and IPA across the box; dashed lines mark the headgroups (vertical) and bulk values (horizontal). (B) Relative mass fraction f_{local}/f_{bulk} vs distance from the headgroup, folded over both leaflets, where f is each species' mass fraction within the solvent. >1 = accumulation, <1 = depletion, showing how water, PG, and IPA compete for the interface.

Results



Conclusions

- IPA and PG both work to disrupt the lipid barrier primarily by increasing the interfacial width**
 - Two effects increase the interfacial width:
 - IPA: disruption of the upper tail region
 - Dominant at low PG - concentrations.
 - PG: dehydration of the headgroup-region
 - Dominant at low or no-water concentration (high PG - concentration)
- The two PEs counter act each other while water is present**
 - Minimum disruption at 30 %_{w/w} Water, 20 %_{w/w} IPA and 50 %_{w/w} PG
 - Cooperative, enhanced disruption in the water free system.
- In the 30 %_{w/w} Water, 20 %_{w/w} IPA and 50 %_{w/w} PG system**
 - PG curbs IPA's penetration
 - Water aggregates around the headgroups and keeps the solvent-lipid interface relatively sharp.

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