

CO₂ adsorption and separation behavior of azole-based ionic liquid impregnated adsorbents supported on porous silica

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Abstract

CO₂ capture from natural gas combustion flue gas is gaining increasing attention. Compared with coal-fired flue gas, it features lower CO₂ concentrations and higher oxygen content, posing challenges for energy-efficient separation. Ionic liquids (ILs) have emerged as promising materials for CO₂ capture due to their low volatility, structural tunability, and low heat capacity. Among them, azole-based ILs exhibit small viscosity change upon CO₂ absorption, enabling stable adsorption/desorption kinetics, especially when immobilized on solid supports. In this study, we prepared solid adsorbents by impregnating tetraalkylphosphonium azolates onto mesoporous silica and investigated their CO₂ capture behavior. The effects of IL structure, loading amount, and support particle size were examined. Photographs of the IL-impregnated adsorbents, experimental setups, and chemical structures of the ILs used are shown in Fig. 1. CO₂ adsorption/desorption experiments were conducted using a fixed-bed column under simulated flue gas conditions (humidified mixture of 5 vol% CO₂ + 95 vol% N₂ for adsorption, and humidified N₂ for desorption). Compared with a tetraethylenepentamine (TEPA)-impregnated reference, the IL-based adsorbents showed lower saturation capacities but higher effective CO₂ capacity. In multi-cycle experiments (5 min adsorption/desorption cycles), [P₄₄₄₄][2CNPyr], [P₆₆₁₄][2CNPyr], and [P₆₆₁₄][BzIm]-impregnated adsorbents exhibited 83%, 76%, and 13% higher effective CO₂ capacity than TEPA, respectively. Further evaluation under cyclic operation was conducted using a two-column vacuum swing adsorption (VSA) setup. In the presentation, we will discuss how IL loading and particle size influence adsorption performance, and present CO₂ recovery and purity data under various process conditions. These findings provide a basis for discussing the potential applicability of azole-based IL-impregnated adsorbents in energy-efficient CO₂ capture from natural gas flue gas.

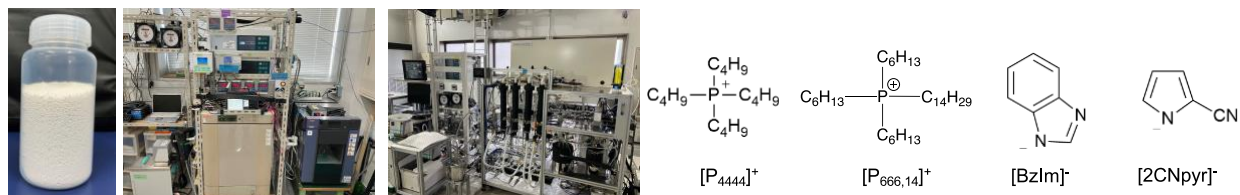


Fig. 1 Photographs of IL-impregnated adsorbent and experimental setups, and chemical structures of ILs used in this study.

Keywords: Ionic liquid; Adsorption; Natural gas combustion flue gas; Vacuum swing adsorption

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