

IEAGHG 8th Post Combustion Capture Conference

16th to 18th September 2025 Marseille, France

Magnesium Extraction from Tailings for CO₂ Mineralization: Insights from Roasting and Acid Leaching Tests

Haftom Weldekidan^{a,*}, Dia Milani^b, Robbie McDonald^a, Phillip Fawell^a, Graeme Puxty^b, Phil Green^b, Paul Feron^c, Nouman Mizra^b

^a CSIRO Mineral Resources, 7 Conlon Street, Waterford, WA 6152, Australia
^bCSIRO Energy Centre, 10 Murray-Dwyer Circuit, Mayfield West, NSW 2304, Australia
^cCSIRO Energy, 36 Gardiner Road, Clayton VIC 3168, Australia

Abstract

Increasing atmospheric concentrations of CO₂ have driven the need for innovative carbon capture and storage strategies. One promising approach is CO₂ mineralization, where magnesium-rich minerals react with CO₂ to form stable carbonate minerals. Mine tailings, a byproduct of mineral processing, can present a valuable feedstock for this purpose when they possess a high magnesium content. This study investigates magnesium extraction from a suitable mine tailings for potential CO2 sequestration. Quantitative X-ray diffraction (QXRD) analysis confirmed the tailings to be primarily composed of 33% serpentine (lizardite) and 20% olivine (forsterite), along with lesser amounts of tale, chlorite (clinochlore), quartz and magnetite. Roasting these tailings with ammonium sulfate at 400°C followed by water leaching enabled 85% magnesium extraction, although this required particles to be <75 µm. In comparison, leaching with sulfuric acid resulted in 90% magnesium extraction, XRD of the acid-leached residue indicated nearcomplete decomposition of the serpentine and olivine phases, confirming the effectiveness of acid leaching in extracting magnesium. The leach residue (primarily quartz and talc) was separated via filtration using a 0.45 µm membrane. Subsequently, Fe, Ni, and Mn were selectively precipitated by gradually increasing the solution pH to 8, leaving a magnesium-rich solution. The extracted magnesium was precipitated as Mg(OH)₂ at pH > 8 then employed for carbonation using pure CO₂, obtained from an amine-based direct air capture (DAC) system. Carbonation of the Mg(OH)₂ was modelled using Aspen plus at both 400°C and ambient temperature, indicating that carbonation at 400°C will result in pure MgCO3 while the carbonation under ambient conditions will favour the formation of hydrated magnesium carbonates. This work demonstrates a viable pathway for integrating the mineral carbonation of tailings with DAC-based CO2, contributing to sustainable waste management and emissions reductions.

Keywords: Tailings, mineral carbonation, direct air capture, serpentine, olivine, CO2 sequestration, acid leaching, roasting

^{*} Corresponding author. Tel.: +61 (08) 9334 8000 E-mail address: Haftom.weldekidan@csiro.au