



Norwegian University of
Science and Technology

The progress in
solvent-based CO₂
capture in the last
decade



Piloting and demonstration

Electricity and heat production

- Hard coal
- Biomass
- Lignite
- Natural gas

Synthetic gas / propane

Industrial

- Cement Kiln
- Steel/Blast furnace
- Silicon production

Waste-to-Energy

Piloting and demonstration

Measurement of process performance (SRD)

Operability and long-term performance



International Journal of Greenhouse Gas
Control

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Conclusions from 3 years of
continuous capture plant operation
without exchange of the AMP/PZ-
based solvent at Niederaussem –
insights into solvent degradation
management

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Roberta Veronezi Figueiredo ^b, Eirini Sk
Juliana Garcia Moretz-Sohn Monteiro ^b



International Journal of Greenhouse Gas
Control

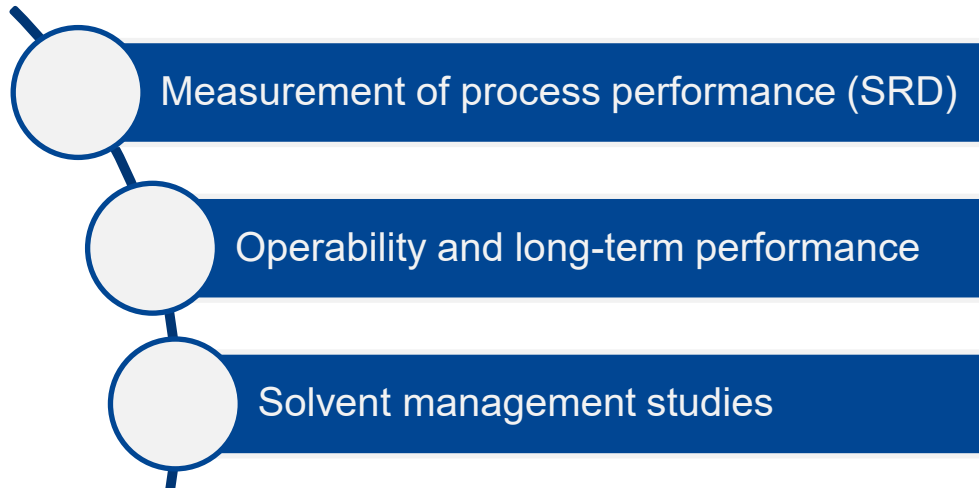
Volume 95, April 2020, 102945



Results of the 18-month test with
MEA at the post-combustion capture
pilot plant at Niederaussem – new
impetus to solvent management,
emissions and dynamic behaviour

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Juliana Garcia Moretz-Sohn Monteiro ^b, Charithea Charalambous ^c, Susana Garcia
^c, Eva Sanchez Fernandez ^c

Piloting and demonstration



Operational issues possible caused by solvent degradation



Foaming



Loss of capture capacity



Increased energy consumption



Corrosion / fouling

Possible environmental / HSE issues



Emission of harmful compounds



Handling of solvent

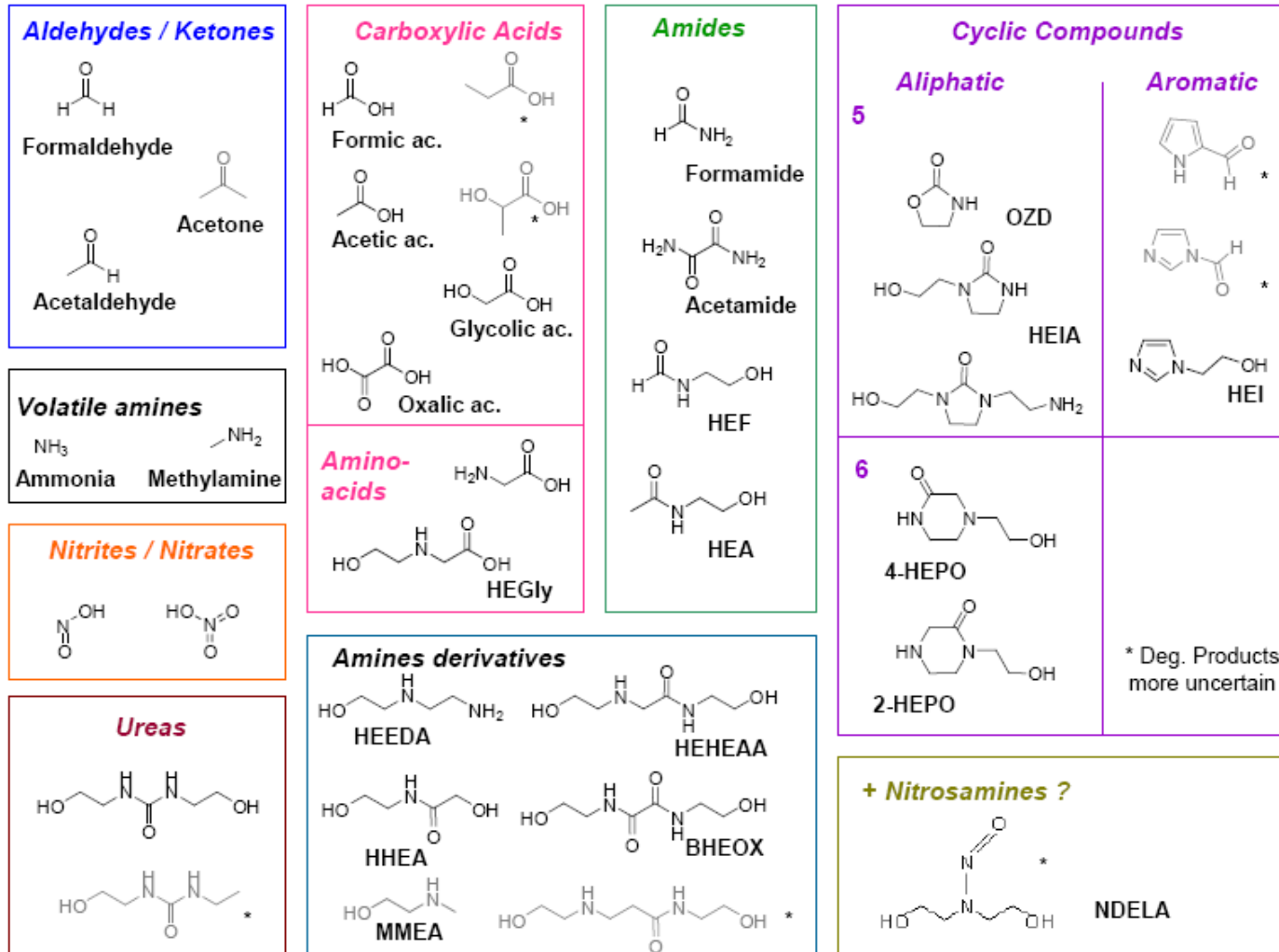


Waste treatment

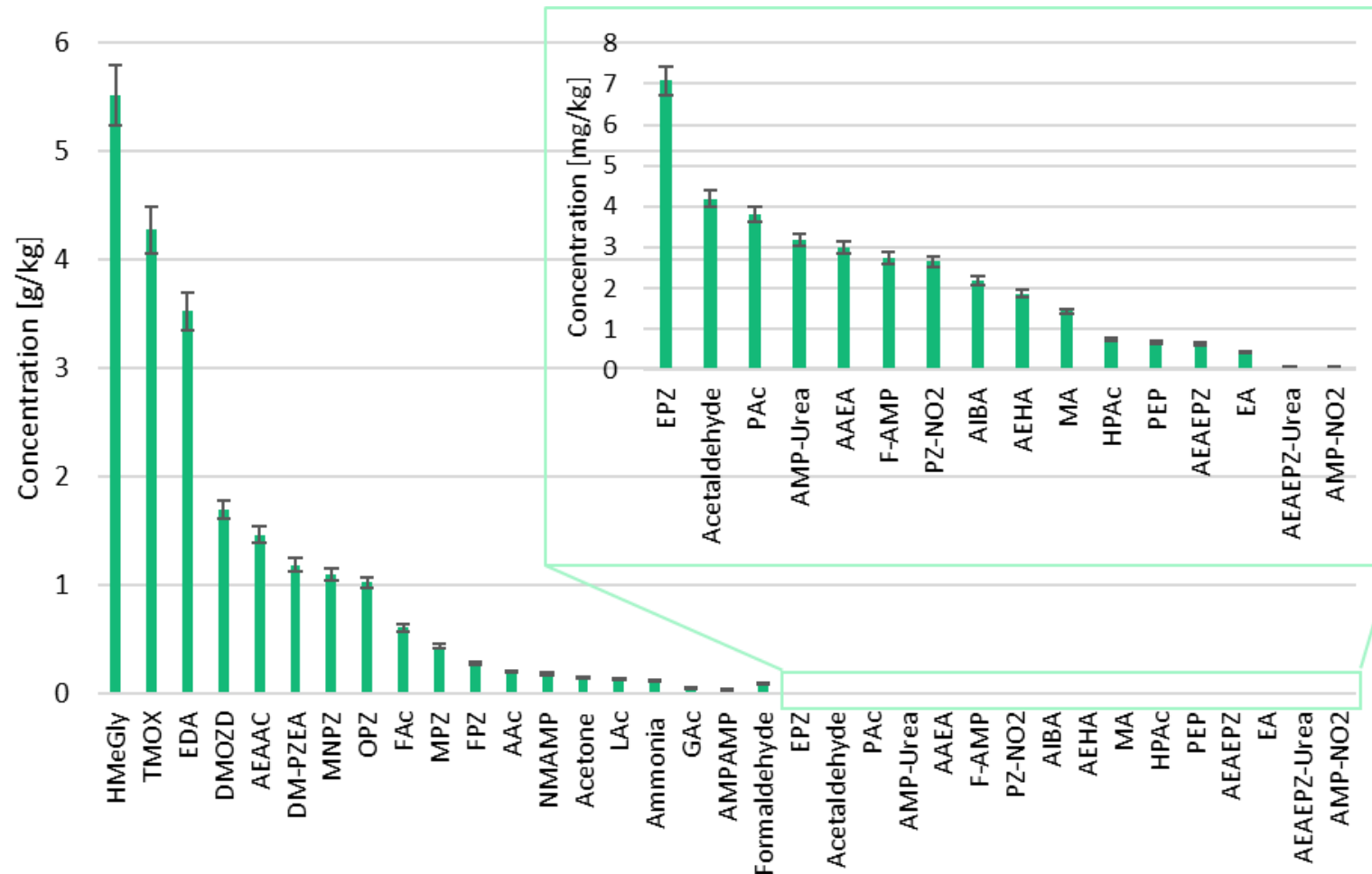


Spill/leakages of harmful compounds

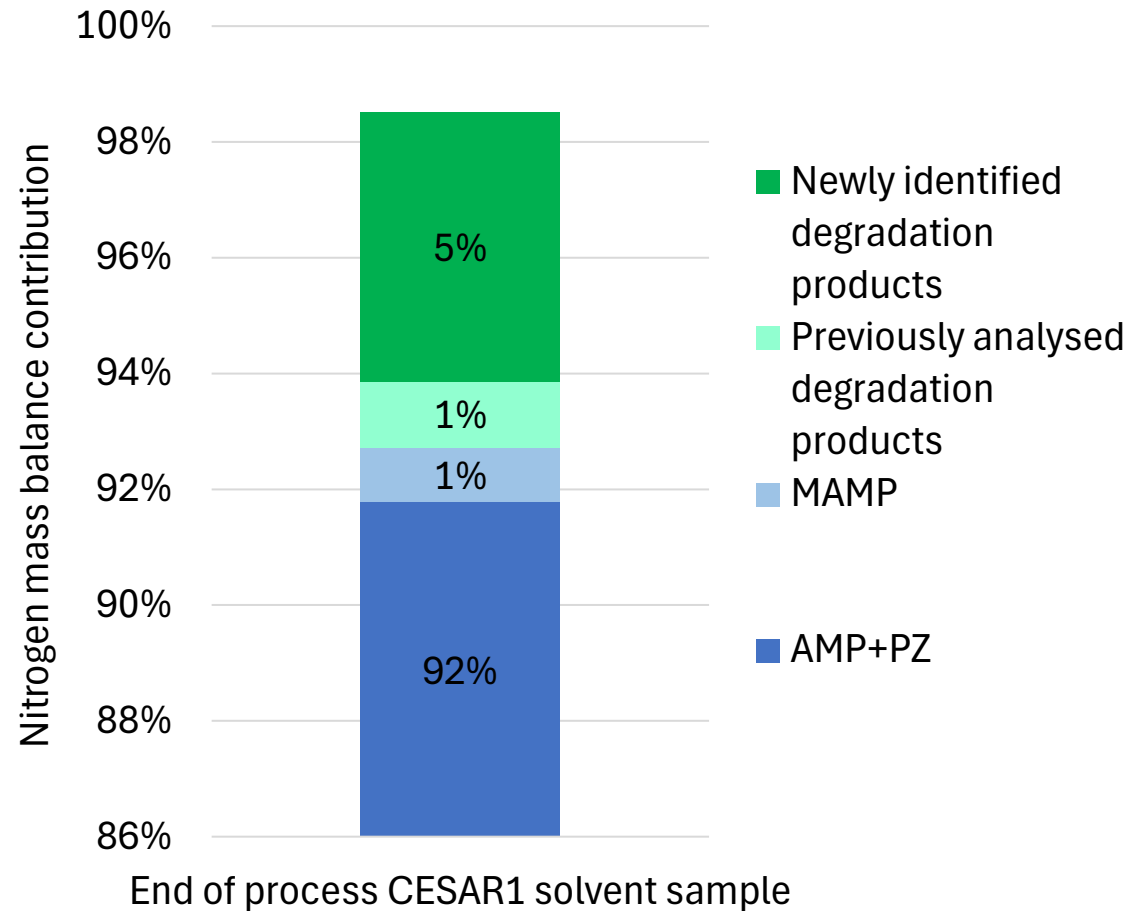
Closing the nitrogen balance



Closing the nitrogen balance

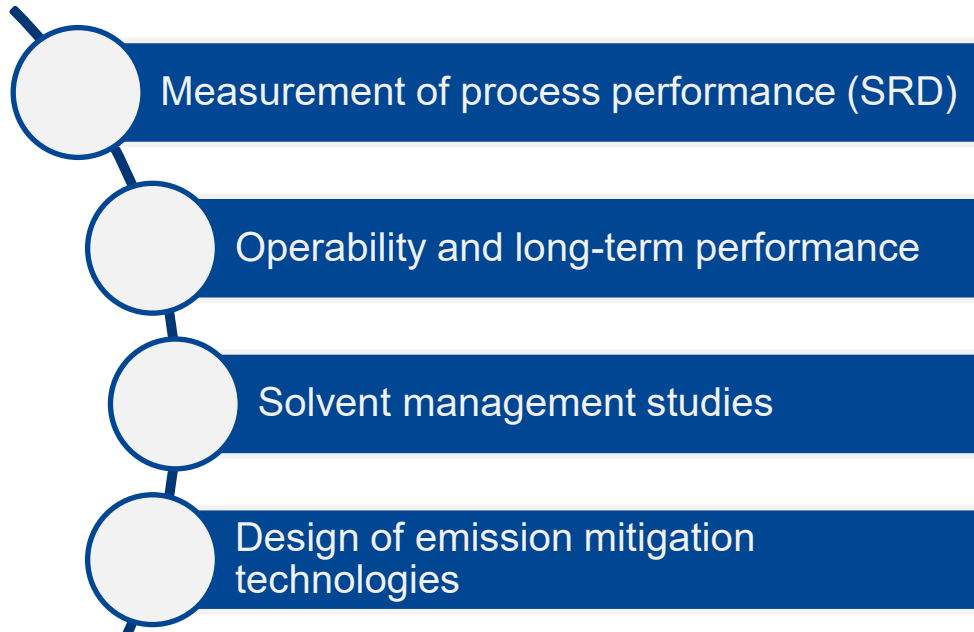


Closing the nitrogen balance



- Optimisation of operational conditions
- Flue gas pretreatment
- Solvent reclamation
- Online monitoring of solvent degradation

Piloting and demonstration

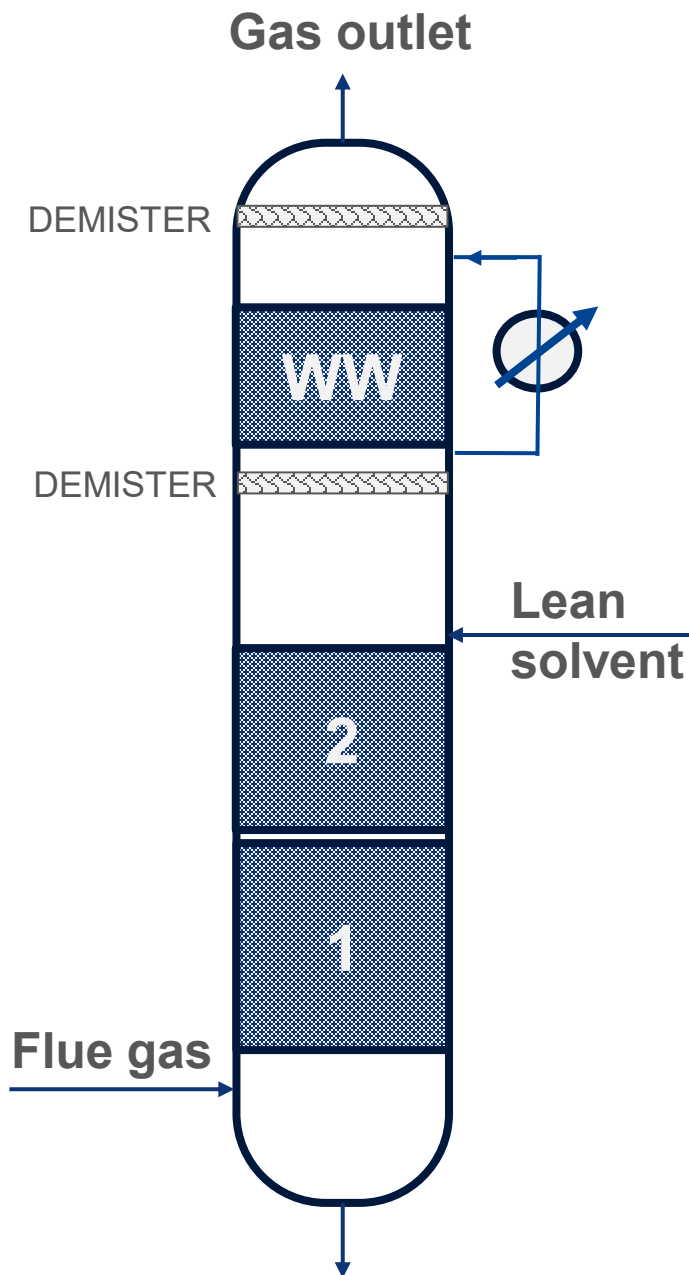


IC, LC-MG, GC-MS, PTR-MS, etc.	FTIR, (LC-MS, PTR-MS etc..)	Solvent amine	Ammonia
		Nitrosamines	Formaldehydes
		Alkylamine	Ketones
		Nitramines	Other?

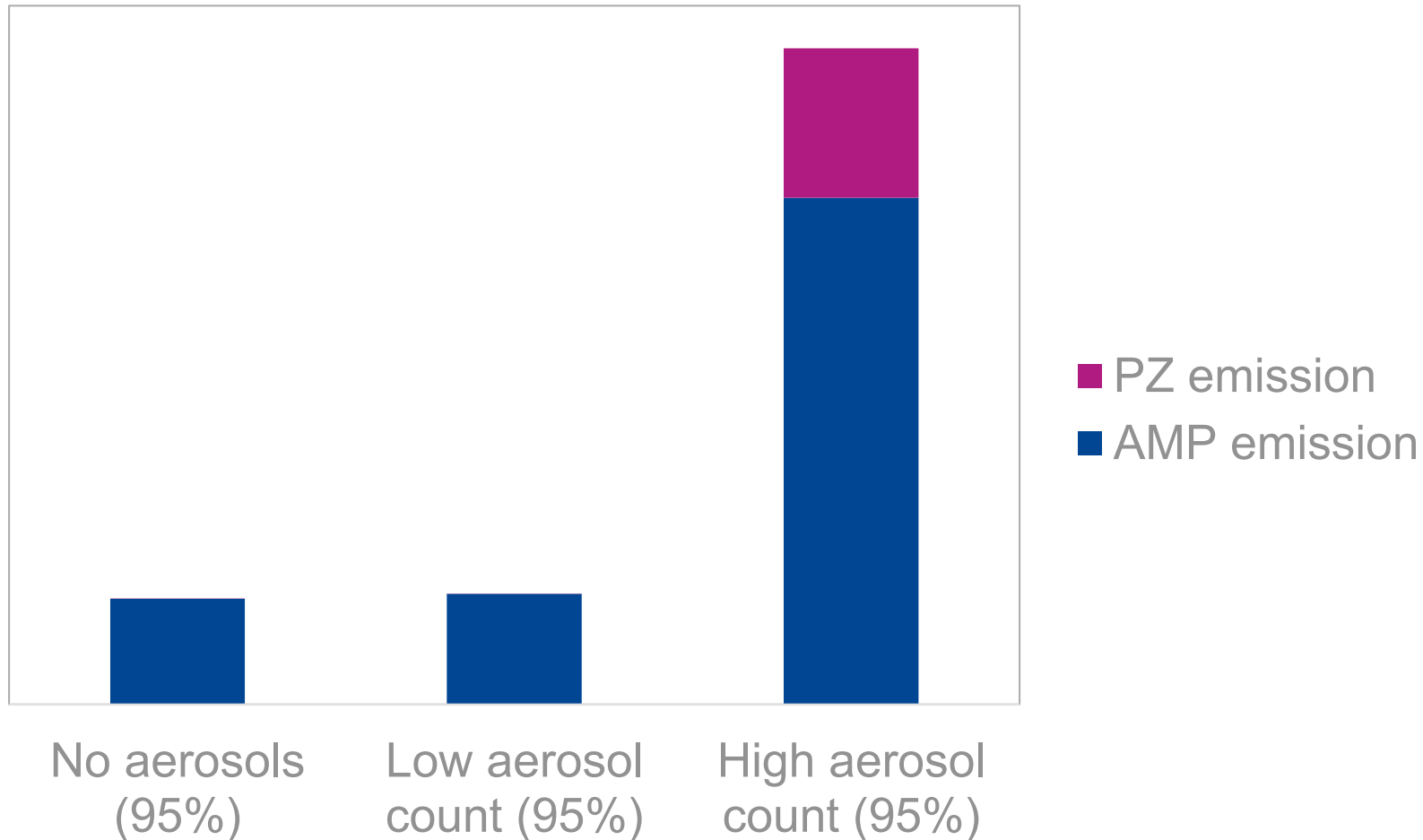
Emissions

Emissions = Volatile + Aerosol + Entrainment

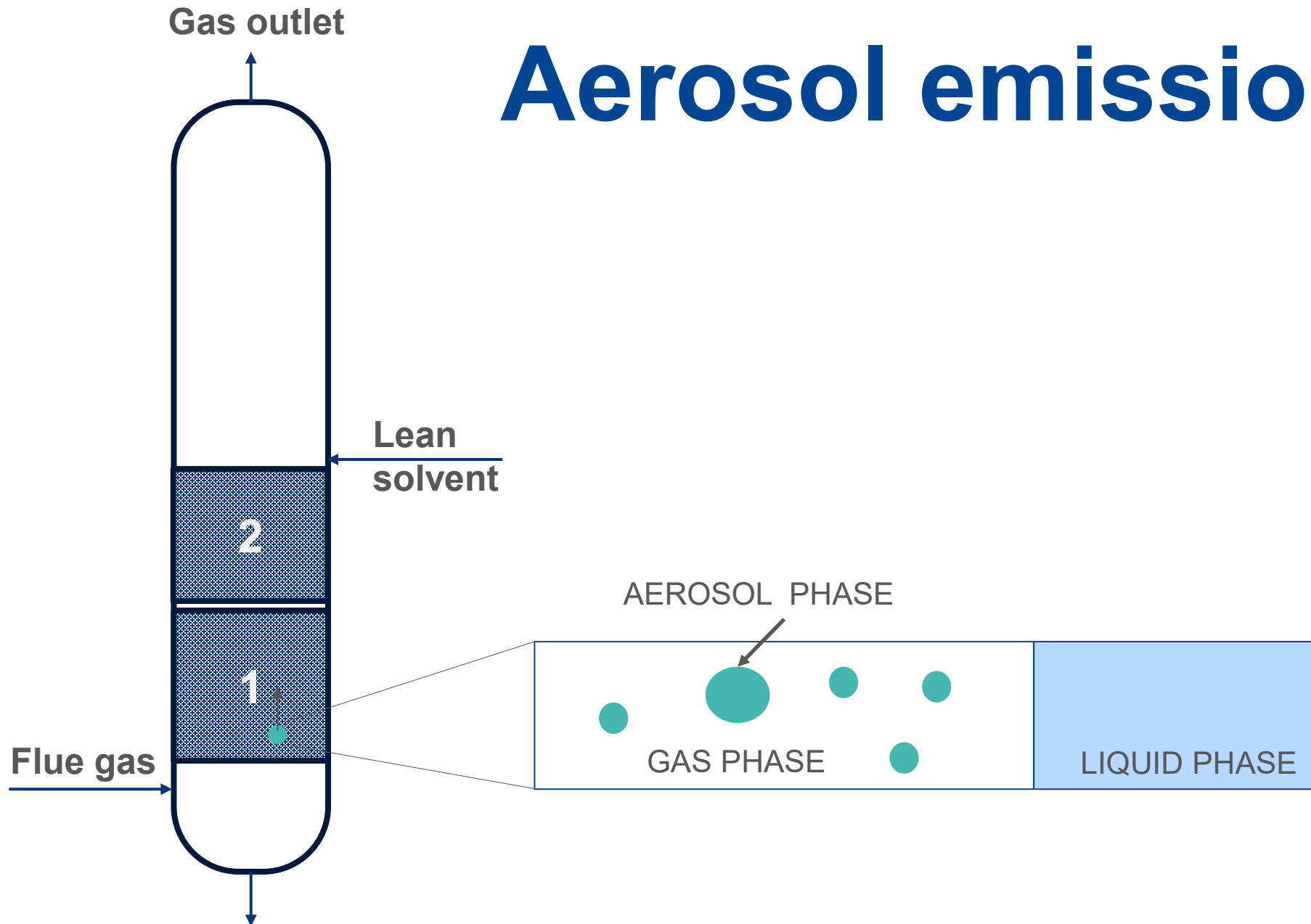
- Solvent amines
- Degradation compounds like ammonia, aldehydes, etc.



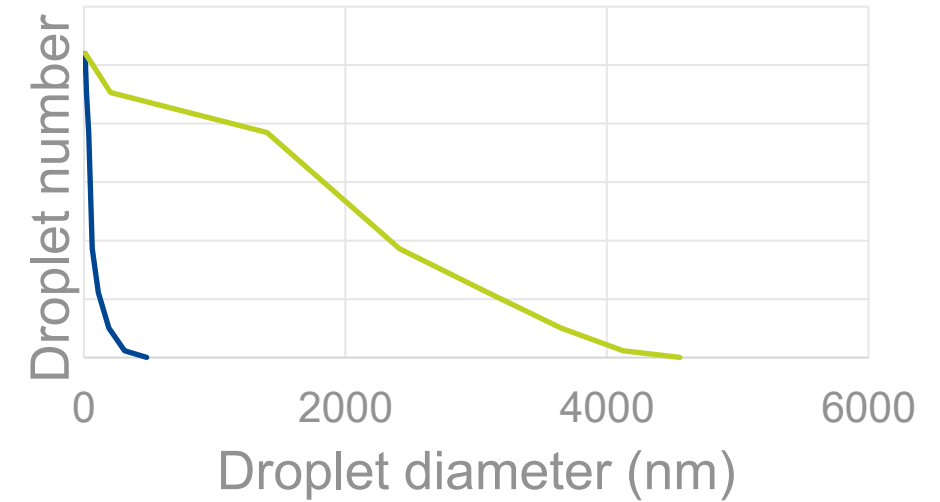
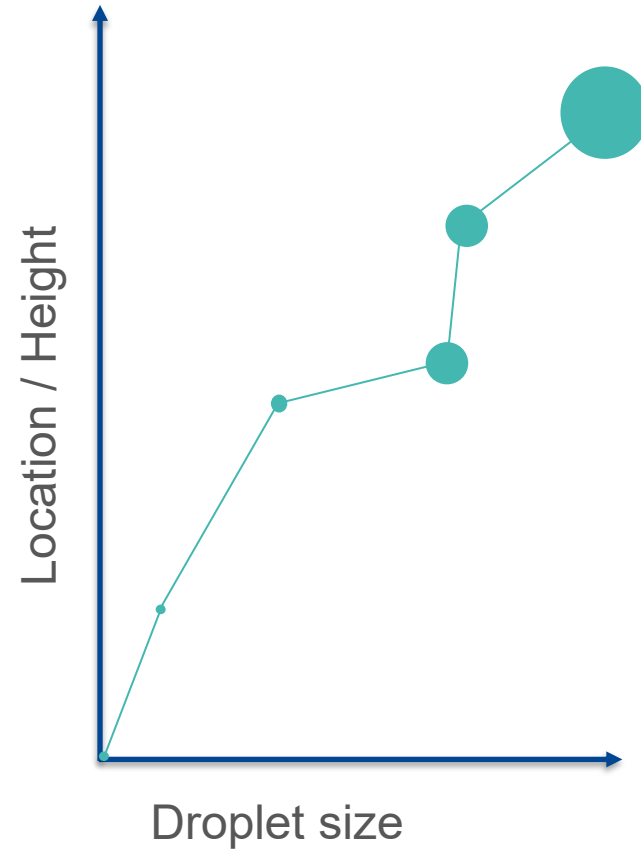
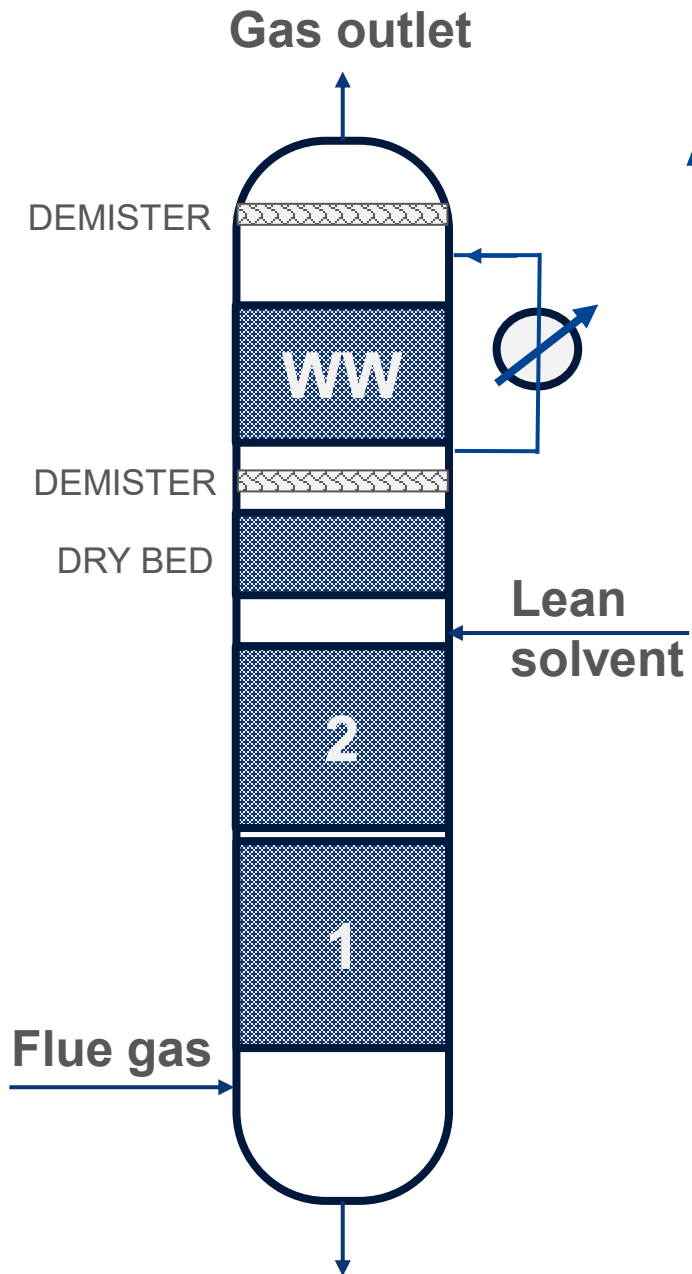
Volatile / aerosol emissions



Aerosol emissions

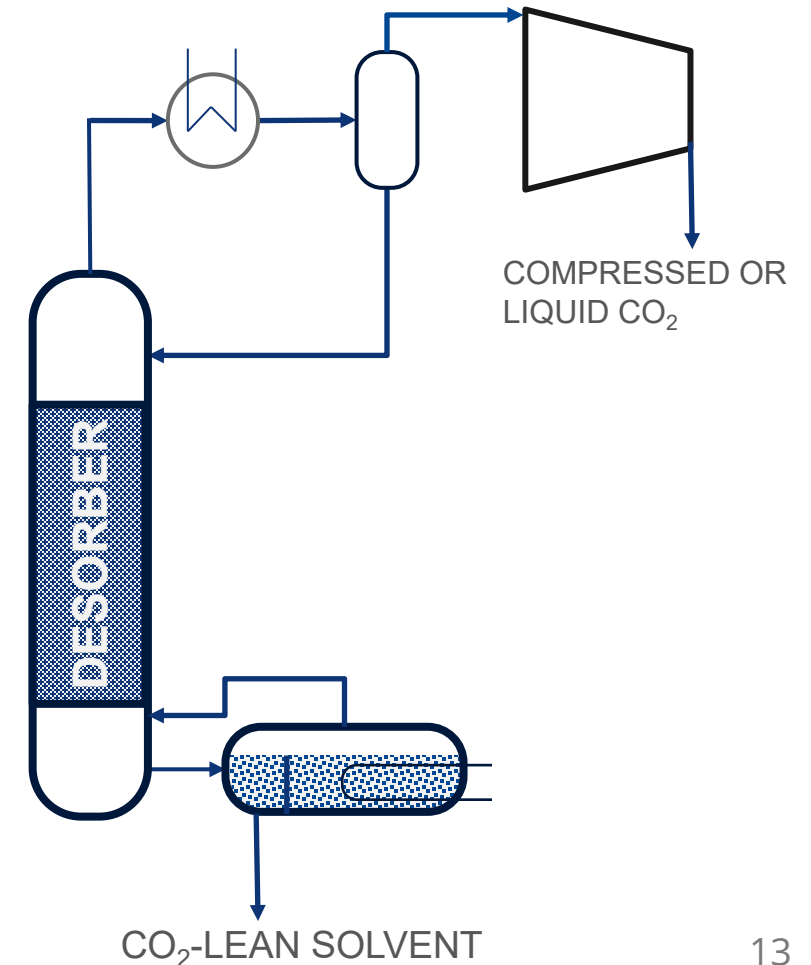


Aerosol emissions



CO₂ purity

- Amines
- Volatile degradation compounds
- Oxygen/Nitrogen
- NO_x?



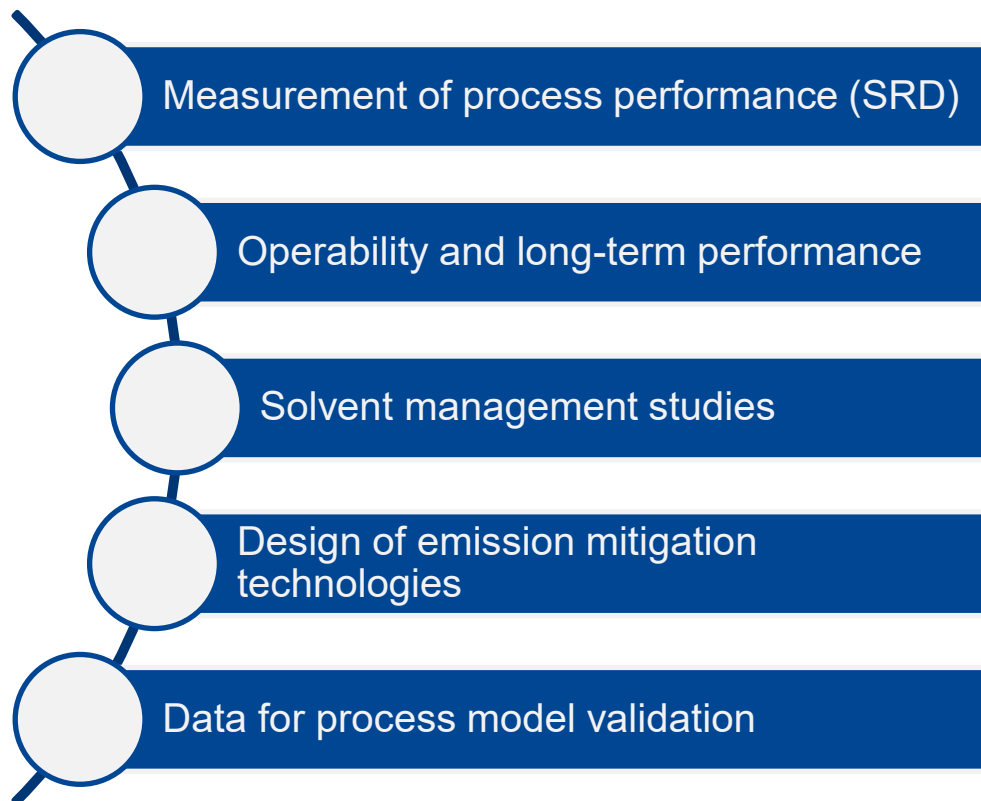
CO₂ purity

- Amines
- Volatile degradation compounds
- Oxygen
- NO_x?
- Online measurements

Table 3. Range and average value measured components sample type 2

Component measured	Range (ppm)	Average (ppm)	Component reference	Limit (ppm)
Sum BTX	0	0	Aromatic hydrocarbon	0.02
Sum CH	0-1.5	0.78	Total volatile hydrocarbons (calc as CH ₄)	50
Sum non-methane CH	0.2-1	0.5	Non-volatile organic residue (oil, grease)	5
Acetaldehyde (C ₂ H ₄ O)	0-0.03	0.006	Acetaldehyde	0.2
Ammonia (NH ₃)	0.02-0.04	0.03	Ammonia	2.5
Benzene (C ₆ H ₆)	0	0	Aromatic hydrocarbon	0.02
Carbon monoxide (CO)	0-0.2	0.1	Carbon monoxide	10
Carbonylsulfide (COS)	0	0	Total Sulfur	0.1
Dimethylether (C ₂ H ₆ O_DIM)	0-0.05	0.003	Total volatile hydrocarbons	50
Ethane (C ₂ H ₆)	0.1-0.4	0.18	Total volatile hydrocarbons	50
Ethanol (C ₂ H ₆ O_ETH)	0-0.2	0.049	Total volatile hydrocarbons	50
Hydrogensulfide (H ₂ S)	0.005-0.015	0.011	Total Sulfur	0.1
Methane (CH ₄)	0-0.5	0.25	Total volatile hydrocarbons (calc as CH ₄)	50
Methanol (CH ₃ OH)	0-0.3	0.05	Methanol	10
Moisture (H ₂ O)	0.1-1	0.46	Moisture	20
Nitrogen dioxide (NO ₂)	0.2-0.8	0.45	Oxides of nitrogen (NO/NO ₂)	2.5
Nitrogen monoxide (NO)	0-0.01	0	Oxides of nitrogen (NO/NO ₂)	2.5
Oxygen (O ₂)	0-30	5.4	Oxygen	30
Propane (C ₃ H ₈)	0-0.08	0.02	Total volatile hydrocarbons (calc as CH ₄)	50
Sulfur dioxide (SO ₂)	0-0.01	0.0015	Total sulfur (as S)	0.1
Toluene (C ₇ H ₈)	0-0.004	0.00016	Aromatic hydrocarbon	0.02
Total Sulfur (TSC)	0-0.002	2,76E-05	Total sulfur (as S)	0.1
Xylene (C ₈ H ₁₀)	0-0.002	0,00013	Aromatic hydrocarbon	0.02

Modelling and simulations



- Rate-based models are the standard
- Models predicting the behaviour of aerosol growth and emissions
- Models describing solvent degradation as part of the process simulations

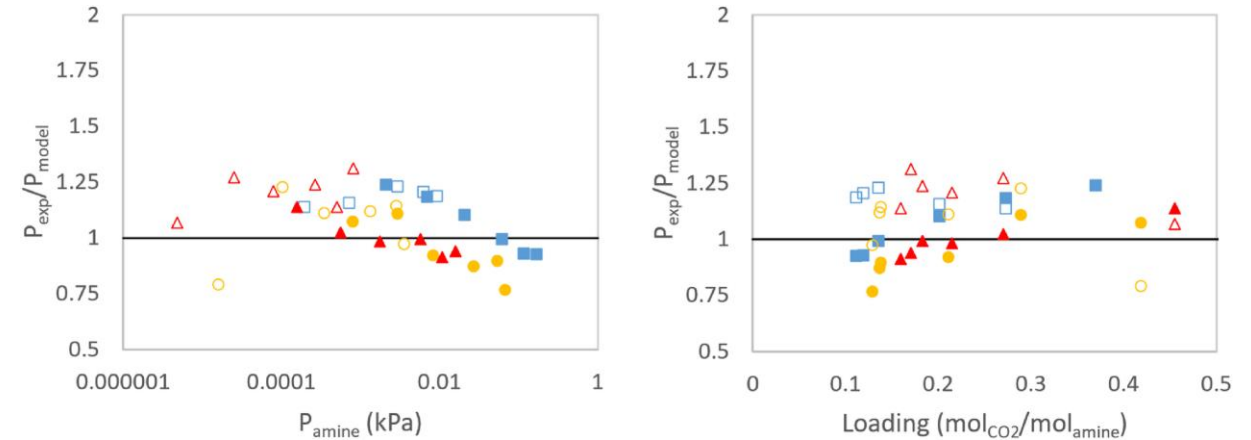
Experimental solvent characterization

The heat of absorption of CO₂

Reaction kinetics and solvent viscosity

CO₂ vapor-liquid equilibrium data

Volatility experiments



Left: Difference between experimental and predicted AMP/PZ partial pressure as a function of the experimental partial pressure; **Right:** Difference between experimental and predicted AMP/PZ partial pressure as a function of loading.

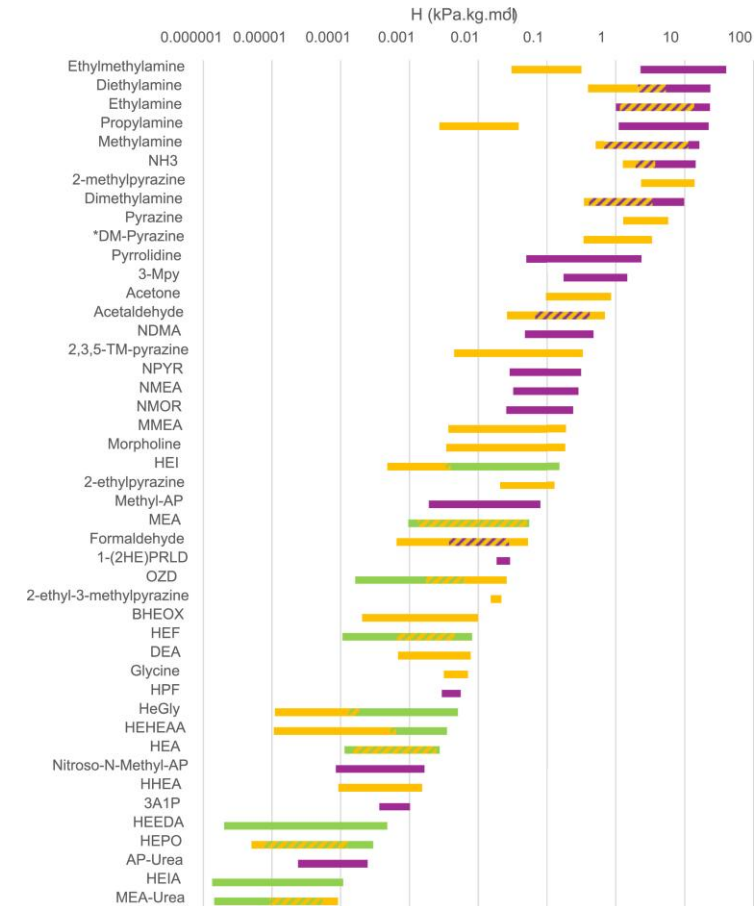
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Concluding remarks

- High quality pilot data
- Testing of new emission and degradation mitigation technologies
- Studies focusing on CO₂ purity
- High quality laboratory data to support model development and to create fundamental understanding
- Analytical and online monitoring methods to identify and monitor degradation compounds
- Development of new solvents

THANK YOU!