





# Design and optimisation of Intensified rotating packed bed (RPB) absorbers for carbon capture from natural gas power plants

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## **Outline**

- 1. Introduction
- 2. Model development, validation & scale-up
- 3. RPB absorber designs options and performance analysis
- 4. RPB absorber intercooling and pressure drop analysis
- 5. Conclusions





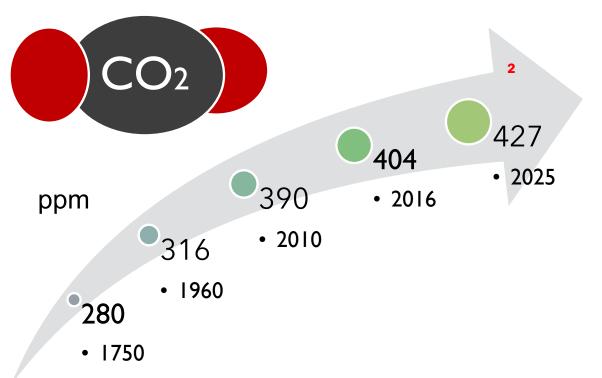
## 1. Introduction

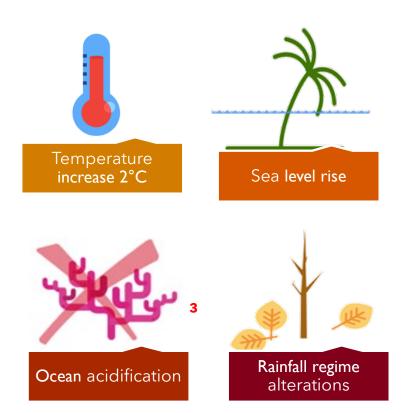
## TotalEnergies

PCCC-8

#### 1.1 Background

- Natural gas combined cycle (NGCC) power plants cleaner than coal but still CO<sub>2</sub> intensive
  - CO<sub>2</sub> from NGCC: 350-450 gCO<sub>2</sub>/kWh<sup>1</sup>





<sup>1</sup>Otitojú O, Oko E, Wang M. Technical and economic performance assessment of post-combustion carbon capture using piperazine for large scale natural gas combined cycle power plants through process simulation. *Appl Energy* 2021;292: 116893 <sup>2</sup>CO<sub>2</sub>

Atmospheric CO<sub>2</sub> concentration for July 2023. Available at: https://www.co2.earth/earths-co2-main-page (Accessed: 08 September 2025 Intergovernmental Panel on Climate Change (2014) Climate Change 2014: Synthesis Report, 5th Assessment Report: Mitigation of Climate doi: 10.1017/CBO9781107415324.



## 1. Introduction

1.1 Background



- Carbon capture essential for net-zero targets
- Carbon capture process based on packed column absorbers and strippers:
  - $\bullet$  remains the industrial benchmark for  $\text{CO}_2$  capture
  - large, costly and high energy penalties
- Rotating Packed Beds (RPBs):
  - offer a step-change in gas-liquid mass transfer through high centrifugal fields
  - reduces equipment footprint
  - enhances efficiency

Wang, M., Joel, A.S., Ramshaw, C., Eimer, D., N. M. Musa (2015), Process intensification for post-combustion  $CO_2$  capture based on Chemical Absorption: a critical review, *Applied Energy*, Vol. 158, p275 – 291. *Highly Cited Paper in Web of Science* 

## Smaller footprint

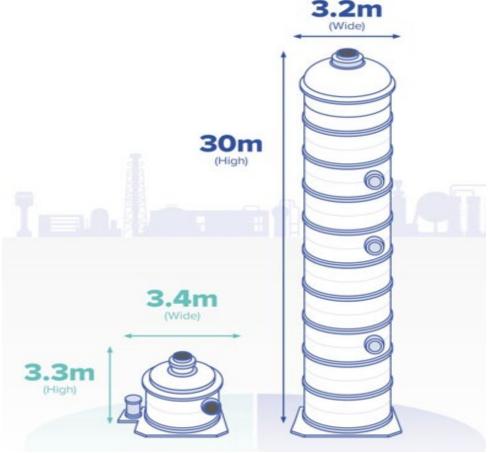


Fig 1. RPB vs Packed column





## 1. Introduction

#### 1.2 Aim and objectives



### Aim

To **design** and **optimise** an intensified post-combustion carbon capture system using various RPB absorber configurations with the **goal of minimising equipment footprint, energy use and cost** associated with carbon capture in natural gas power plants.

## **Objectives**

- Model development, validation and scale-up of RPB absorber.
- Design of different RPB absorber configurations (i.e. single, multiple, series and parallel arrangements).
- Implementation of RPB absorber intercooling.
- Process optimisation of the intercooled RPB absorbers





Otitoju, O., Oko, E., Wang, M. (2023), Modelling, scale-up and techno-economic assessments of rotating packed bed absorber for CO<sub>2</sub> capture from a 250 MWe combined cycle gas turbine power plant, *Applied Energy*, Vol. 335, 120747.

## 1. Introduction

#### 1.3 Research Methodology



#### RPB absorber

- Model development
- Model validation
- Model scale-up

#### RPB absorber designs & performance analysis

- Option 1–One RPB absorber at 55 wt% and 75 wt% MEA
- Option 2–Two RPB absorbers in parallel at 75 wt% MEA
- Option 3– EGR with One RPB absorber at 75 wt% MEA
- Option 4– EGR with Two RPB absorbers in parallel at 75 wt% MEA

#### RPB absorber intercooling & process analysis

- Absorber intercooling
- Pressure drop analysis

#### Process optimization & cost

- Optimal process conditions
- Carbon capture cost







## University of Sheffield | Chemical & Biological Engineering | Chemical & Sheffield | Chemic



#### Material balance for the gas phase

$$0 = -\frac{1}{2\pi rZ} \frac{\delta(F_g y_i)}{\delta r} - a_{gl} N_i \tag{1}$$

#### Material balance for the liquid phase

$$0 = \frac{1}{2\pi rZ} \frac{\delta(F_l x_i)}{\delta r} + a_{gl} N_i \tag{2}$$

#### **Energy balance for the gas phase**

$$0 = -\frac{F_g C_{p,g}}{2\pi r Z} \frac{\delta T_g}{\delta r} + a_{gl} h_{gl} (T_l - T_g)$$
(3)

#### **Energy balance for the liquid phase**

$$0 = \frac{F_l C_{p,l}}{2\pi r Z} \frac{\delta T_l}{\delta r} - a_{gl} \left( h_{gl} (T_l - T_g) - \Delta H_{rxn} N_{CO2} - \Delta H_{vap,H_2O} N_{H_2O} - \Delta H_{vap,MEA} N_{MEA} \right)$$
(4)

Model developed in *Aspen Custom Modeler V11* with eNRTL-RK method.





## 2. Model development, validation & scale-up



2.2 Model Validation

Validated against pilot-scale data from Jassim et al<sup>4</sup>.

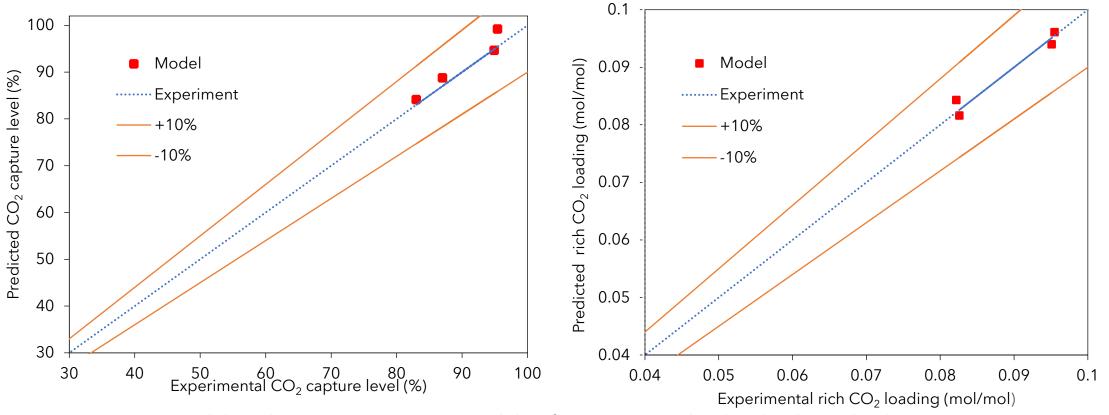


Fig 3. Model predictions versus experimental data for CO<sub>2</sub> capture level and Rich CO<sub>2</sub> loading

Model predictions align with experimental data for CO<sub>2</sub> capture levels and rich solvent
 CO<sub>2</sub> loading

<sup>4</sup>Jassim MS, Rochelle G, Eimer D, Ramshaw C. Carbon dioxide absorption and desorption in aqueous monoethanolamine solutions in a rotating packed bed. *Ind Eng Chem Res* 2007;46:2823-33.





## 2. Model development, validation & scale-up



Validated RPB model was scaled up to assess designs for >100 MW CCGT power plant.

#### Lean solvent flow rate (L)

$$L = \frac{Fx_{CO_2}\Psi_{CO_2}}{100\check{\mathsf{Z}}\Delta\alpha} \left[ \frac{M_{MEA}}{44.009} \left( 1 + \frac{1 - \omega_{MEA}}{\omega_{MEA}} \right) + \check{\mathsf{Z}}\alpha_{lean} \right]$$

 $M_{MEA}$  is the molar mass of MEA  $\Delta\alpha$  (delta alpha) is the difference between rich and lean CO<sub>2</sub> loading  $(\alpha_{rich} - \alpha_{lean}).$ 

 $W_{MEA}$  is the wt% of MEA

 $x_{CO_2}$  is the mass fraction of  $CO_2$  (wt%)

 $\Psi_{CO_2}$  is the desired CO<sub>2</sub> capture level (%)

F is the flue gas mass flow rate (kg/s)

## The inner radius $(R_i)$

$$R_i = \left(\frac{G}{\pi v_{iet} (1 - f_d)}\right)^{\frac{1}{2}} \left(\frac{4\rho_G}{\rho_L}\right)^{\frac{1}{4}}$$

Where:

G = Volumetric gas flow rate, m<sup>3</sup>/s

 $v_{iet}$  = Liquid jet velocity (4-5 m/s recommended)

 $f_d$  = Fraction of the packing inner radius that the liquid distributor occupies about 0.25-0.3)

 $\rho_G$  = Gas phase density (kg/m<sup>3</sup>)

 $\rho_L$  = Liquid phase density (kg/m<sup>3</sup>) (increases with MEA concentration thereby resulting in lesser r<sub>i</sub>)

#### The axial height (Z)

The gas phase superficial velocity (U<sub>G</sub>) can be obtained using the Sherwood flood correlation (Singh et al. 1992);

$$\log \frac{U_{G_{flood}}^2 a_t}{R_i \omega^2 \varepsilon^3} \left[ \frac{\rho_G}{\rho_L} \right] \left[ \frac{\mu_L}{\mu_W} \right]^{0.2} = -2.27 - 1.14 \log \frac{L}{G} \sqrt{\frac{\rho_G}{\rho_L}} - 0.17 \left[ \log \frac{L}{G} \sqrt{\frac{\rho_G}{\rho_L}} \right]^2$$

Axial height (Z) is obtained as follows (Agarwal et al., 2010)

$$Z = \frac{G}{2\pi R_i U_G}$$

#### The Outer radius (R<sub>o</sub>)

The outer diameter  $(R_0)$ Finally, the outer diameter is calculated:

$$\pi(R_o^2 - R_i^2) = ATU \times NTU$$

• Agarwal, L., Pavani, V., Rao, D.P. and Kaistha, N. Process Intensification in HiGee Absorption and Distillation: Design Procedure and Applications. Ind. Eng. Chem. Res. 2010, 49, 10046–10058. PCCC-8



## Chemical & 2. Model development, validation & scale-up

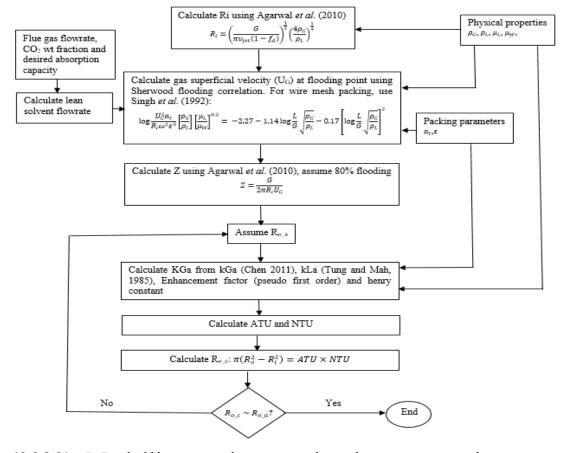
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2.3 Scale-up

■ Flue gas at nearly 500°C cooled to ~40.5°C via direct contact

cooler (DCC).



Otitoju, O., Oko, E., Wang, M. (2023), Modelling, scale-up and techno-economic assessments of rotating packed bed absorber for CO<sub>2</sub> capture from a 250 MWe combined cycle gas turbine power plant, *Applied Energy*, Vol. 335, 120747.





TotalEnergies

3.1 Option 1 - One RPB Absorner (55 wt% MEA vs 75 wt% MEA)

■ In Design Option 1, a single RPB absorber is used, operating with 55 wt% MEA and 75 wt% MEA for a large-scale CCGT power plant (>100 MW).

Table 2: RPB absorber size and operating conditions for design option 1

	MEA concentrations		
	55 wt%	75 wt%	
R <sub>i</sub> (m)	***	***	
$R_o(m)$	***	***	
Z <sup>°</sup> (m)	***	***	
Lean flow rate (kg/s)	***	***	
L/G ratio (kg/kg)	0.67	0.53	
Lean Temperature (K)	313.15	313.15	
RPB Pressure (bar)	1.01325		
Rotation (RPM)	200		
Packing type	Expamet		
Surface area (m²/m³)	2,132		





3.1 Option 1 - One RPB Absorner (55 wt% MEA vs 75 wt% MEA)



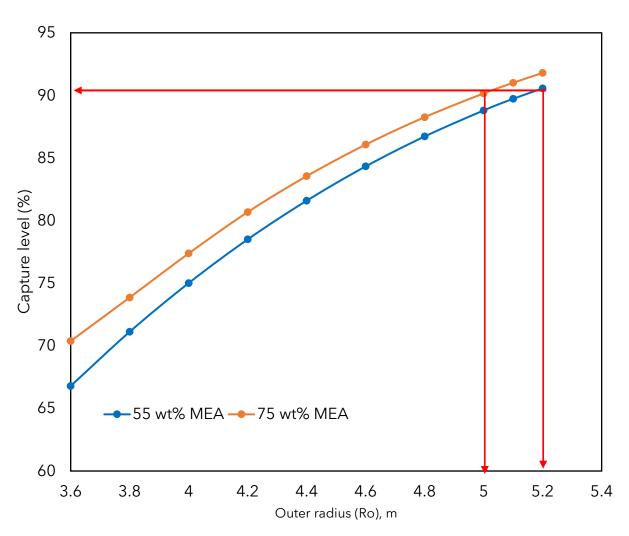


Fig 4. influence of outer radius on capture level

- With design Option 1.
   Key findings include;
- Achieved 90% CO<sub>2</sub> capture.
- 75 wt% MEA reduces size by roughly 5%
- Rotor power: 3,240 kW at 55 wt% MEA vs 1,995 kW at 75 wt% MEA.
- 38% less rotor energy consumption with 75 wt% MEA because of lower solvent flow and RPB size.





TotalEnergies

3.2 Option 2 - Two RPB Absorbers in parallel (75 wt% MEA)

With two RPB absorbers in parallel, the flue gas flow rate is halved in a splitter.

Table 3: RPB absorber size and operating conditions for design Option 2

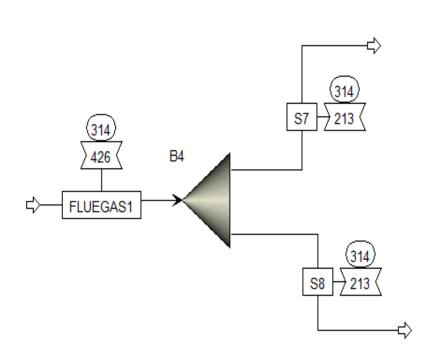


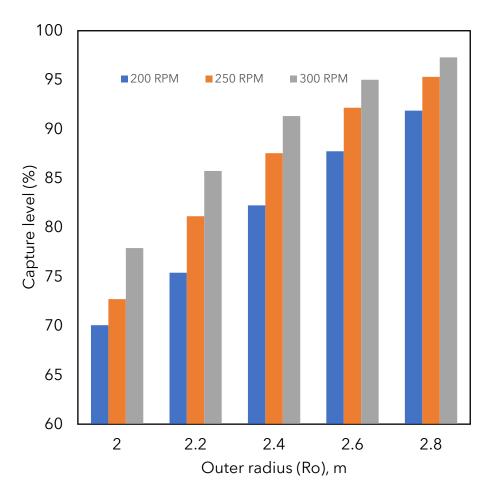
Fig 5. Flue gas stream to each of the RPB absorber

RPB	MEA conc. 75 wt%	
No of RPB	2	
R <sub>i</sub> (m)	***	
$R_o(m)$	***	
Z (m)	***	
Lean flow rate (kg/s)	***	
Lean Temperature (K)	313.15	
RPB Pressure (bar)	1.01325	
Rotation (RPM)	200-300	
Packing type	Expamet	
Surface area (m²/m³)	2,132	



3.2 Option 2 - Two RPB Absorbers in parallel (75 wt% MEA)





**Fig. 6** effect of outer radius on capture level across various rotor speeds

#### **Key findings for Option 2**

Increasing rotor speed **reduces the required absorber size** (outer radius, Ro) to achieve 90% CO<sub>2</sub> capture. However, it significantly **increases rotor energy consumption**.

#### At 200 RPM:

Rotor power = 652.52 kW for two RPBs

#### **At 250 RPM:**

Rotor power = 873.88 kW for two RPBs

#### **At 300 RPM**:

Rotor power = 1111.64 kW for two RPBs Rotor energy at 200 RPM is **41.3% lower** than at 300 RPM.

**Trade-off:** Higher speed improves compactness but at the cost of greater energy use.





### University of | Chemical & 3. RPB absorber design options & Performance analysis

Sheffield Biological Engineering 3.3 Option 3 - One RPB Absorber+ EGR (75 wt% MEA)

■ EGR of 35.44% produced 6 vol %  $CO_2$  composition in the flue gas.

✓ EGR was varied from 0 - 50 vol%

**Table 4.** Exhaust gas flowrate and composition at 35.44% EGR and composition and flow rate after cooling (*Confidential*)





## University of Sheffield | Chemica Biologic Enginee | 3. RPB absorber design options & Performance analysis | 3.4 Option 4 - Two RPB Absorbers + EGR (75 wt% MEA)



With Option 4 - (2 RPB absorbers in parallel + EGR + 75 wt% MEA), the flue gas flow rate is halved in a splitter.

## **Key Findings for Option 4**

- 15% less R<sub>o</sub> required to achieve 90% capture at 300 RPM compared to at 200 RPM
- Rotor speed increase reduces absorber size (Ro) for achieving 90% CO<sub>2</sub> capture:
- However, it significantly increases rotor energy consumption: 301.02 kW at 200 RPM vs 489.06 kW at 300 RPM
- Operating at 200 RPM saves 38.5% energy compared to 300 RPM.
- Insight: A higher rotor speed results in a more compact design but at the expense of much higher energy usage.





### 3.5 Comparison 4 design options



- For the 4 design options compared.
  - EGR, MEA concentration and rotor speed key to performance.
  - EGR lowers flue gas flow.
  - Option 4 at 250 RPM:
  - > 4× smaller RPBs than Option 1
  - Lowest energy
  - ➤ Best size-energy trade-off.
  - ➤ Option 4 (250 RPM) recommended for further analysis





#### 4. RPB absorbers with Inter-cooler & Performance analysis



**4.1 RPB Absorbers with Inter-cooler** 

- Liquid phase temperature rise observed in the RPB
- Intercoolers will lower temperature to enhance absorption in the RPB
- The benefits of intercoolers in RPB absorbers explored.

Oko, E. Wang, M., Ramshaw, C. (2018), Intercooling for rotating packed bed absorbers in intensified solvent-based post-combustion CO<sub>2</sub> capture process, *Applied Energy*, Vol. 223, p302-316.

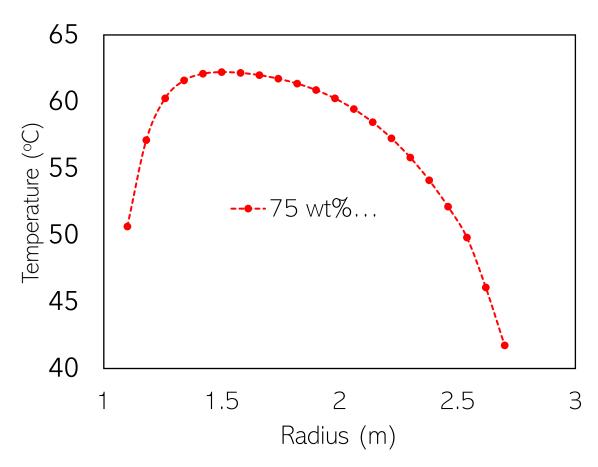


Fig 8. Liquid temperature profile in RPB absorber with 75 wt% MEA





## 4.1 RPB absorber with Intercoolers



## Performance Improvements of RPBs with Intercooler

- 16% reduction in total packing volume compared to a single RPB without intercooling.
- 14% drop in the maximum liquid phase temperature, improving absorption performance by up to 12%.
- Intercooling led to a 5% reduction in solvent flowrate, enabling further operational efficiency.

## **Insights**

 Intercooling is essential in mitigating temperature rise within the RPB absorber, which can otherwise reduce CO₂ capture efficiency.





4.2.1 pressure drop correlations

## **Pressure Drop Correlations Used**

- ➤ Liu et al. (1996)
- Sandilya et al. (2001)
- >Agarwal et al. (2010)
- ➤ Neumann et al. (2017)







4.2.1 pressure drop correlations

## The **pressure drop correlation** by **Liu et al. (1996)** is described by these equations

$$\Delta P = \Delta P_d + \Delta P_w$$

$$\Delta P_{d} = \frac{1}{2} F v^{2} \frac{a_{t}}{\epsilon^{3}} f_{d} (R_{o} - R_{i})$$

$$\Delta P_{w} = \frac{1}{2} F v^{2} \frac{a_{t}}{(\in -\in_{L})^{3}} f_{w}(R_{o} - R_{i})$$

Where,

 $\Delta P_d$  = pressure drop of dry bed (Pa)

 $\Delta P_w$  = pressure drop of wet bed (Pa)

 $F_V$  = gas capacity factor (kg<sup>1/2</sup>/m<sup>1/2</sup> s)

 $a_t$  = specific area of packing per unit volume (m<sup>2</sup>/m<sup>3</sup>)

 $\varepsilon$  is the bed porosity (m<sup>3</sup>/m<sup>3</sup>)

 $f_w$  = Wet resistance factor for cylindrical packing

f<sub>d</sub>=resistant coefficient







### 4.2.1 pressure drop correlations

The pressure drop correlation by **Sandilya et al. (2001)** consists primarily of the following components:

$$\Delta P = \Delta P_c + \Delta P_f + \Delta P_m$$

(i) Centrifugal pressure drop

$$\Delta P_c = \frac{1.8}{8} \rho_g \omega^2 (R_o^2 - R_i^2)$$

(ii) Frictional pressure drop

$$\Delta P_{\rm f} = \frac{(1 - \epsilon)}{\epsilon^3} \frac{Q_g}{2\pi z d_p} \left( \frac{150(1 - \epsilon)\mu_g}{d_p} \ln \frac{R_o}{R_i} \right) + 1.75 \frac{Q_g \rho_g}{2\pi z} \left( \frac{1}{R_i} - \frac{1}{R_o} \right)$$

(iii) Momentum-gain pressure drop

$$\Delta P_m = \frac{1}{2} \rho_g \left( \frac{Q_g}{2\pi z \epsilon} \right)^2 \left( \frac{1}{R_i^2} - \frac{1}{R_o^2} \right)$$







4.2.1 pressure drop correlations

The pressure drop correlations by **Agarwal et al. (2010)** is described by this equation.

$$\begin{split} &\Delta P_{packing} \\ &= \frac{1}{2} \rho_g A \omega^2 \left( R_o^2 - R_i^2 \right) + \frac{f \rho_g}{2 d_h} \left( \frac{Q_g}{2 \pi Z \epsilon} \right)^2 \left( \frac{1}{R_i} - \frac{1}{R_o} \right) \\ &+ \frac{1}{2} \rho_g \left( \frac{Q_g}{2 \pi Z \epsilon} \right)^2 \left( \frac{1}{R_i^2} - \frac{1}{R_o^2} \right) \end{split} \tag{8}$$







4.2.1 pressure drop correlations

The pressure drop correlation by **Neumann et al. (2017)**, is described by the following equations;

$$\begin{split} & \Delta P_{total} \\ &= A_{CH} \frac{\rho_g \omega^2}{2} \left( R_o^2 - R_i^2 \right) + \Psi_0 (1 - \varphi) \frac{(1 - \epsilon)}{\epsilon^3} \frac{F_{G,int}^2}{d_p} (R_o - R_i) \\ &+ \Delta P_{f,empty} + \Delta P_{f,rotor} + \Delta P_{rotor} \ (9) \end{split}$$







4.2.2 Sensitivity analysis and results

## Effect of Ro on pressure drop

• Pressure drop increases with Ro:

> 100%: 7.7-23.7 kPa

> 110%: 8.1-23.1 kPa

> 120%: 9.8-28.1 kPa







4.2.2 Sensitivity analysis and results

## Effect of Z on pressure drop

- Pressure drop reduces as Z increases:
  - > 100%: 8.7-26.2 kPa
  - > 105%: 8.1-26.2 kPa
  - > 110%: 7.5-20.5 kPa
  - > 115%: 6.9-18.3 kPa





## TotalEnergies

4.2.3 comparison of packing types

### Comparison of pressure drop in the RPB

	Expamet packing	Stainless steel wire mesh packing		
Pressure drop	Sandilya et	_	Agarwal et	Neumann et al
correlation	al. (2001)	al. (2001)	al. (2010)	(2017)
Pressure drop (Pa)	19585	9118.52	8704.4	9663.26
Porosity	0.76		0.96	
Surface area				
$(m^2/m^3)$	2132		803	

Wire Mesh: Lower pressure drop ( $\approx 2 \times$  less) compared to Expamet





## 5. Conclusions



- EGR, MEA concentration and rotor speed are key to RPB absorber performance.
- Best performance achieved with the design consisting of 2 RPB absorbers, EGR and 75 wt% MEA i.e. above 90% and smaller RPB size (up to 4 times smaller).
- Porosity and surface area of packing are critical in RPB design.
- Wire mesh could achieve ≈2× less pressure drop compared to Expamet.
- Wire mesh offers optimal balance between efficiency and pressure drop





## 5. Conclusions



### Optimal Operating Conditions Identified

- Rotor Speed: 275-300 RPM offers the best balance between energy consumption and cost.
- L/G Ratio: 0.9-1.0 kg/kg minimizes  $CO_2$  capture cost while maintaining efficiency.
- Reducing MEA solvent make-up by **10-20%** could lower costs significantly (~\$68,000/yr savings).

### Economic Performance Insights

- CO<sub>2</sub> Capture Cost: \$17.52/ton, competitive with alternative capture methods.
- Total Annualized Cost (TAC): \$4.01 million/yr, with major cost contributors being MEA solvent, electricity and O&M.

