



CARDC

China Aerodynamics R&D Center 中国空气动力研究与发展中心

2021 IAA Planetary Defense Conference
April 26-30, 2021, Vienna, Austria

Numerical Analysis of Aerodynamic Heating on Asteroid During Entry to Earth's Atmosphere

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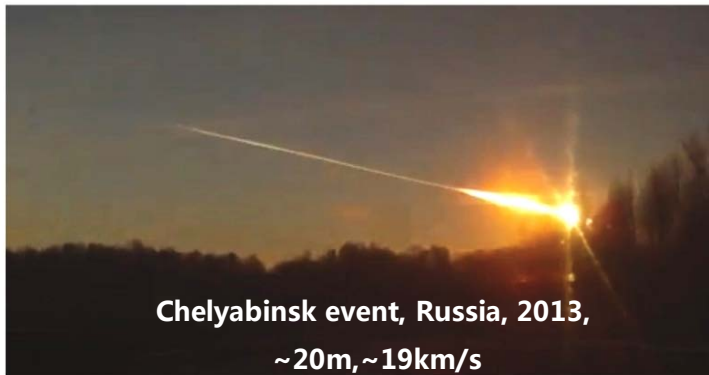
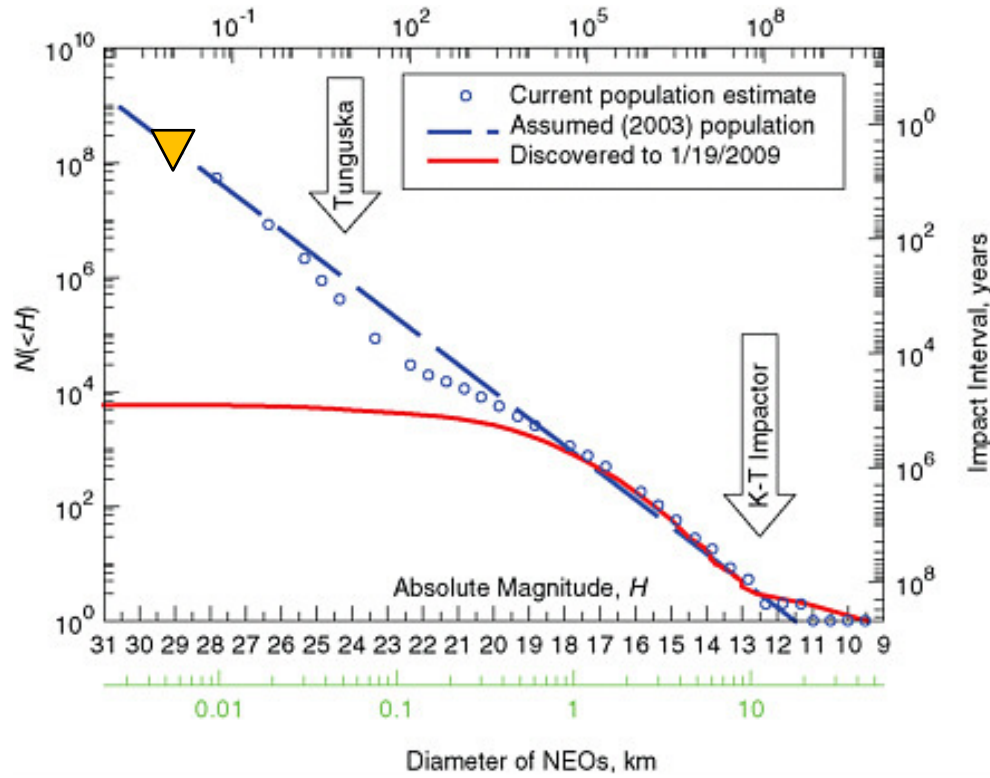
Outline

- 1、 Background
- 2、 Numerical method and physical-chemical model
- 3、 Results and discussion
- 4、 Summary



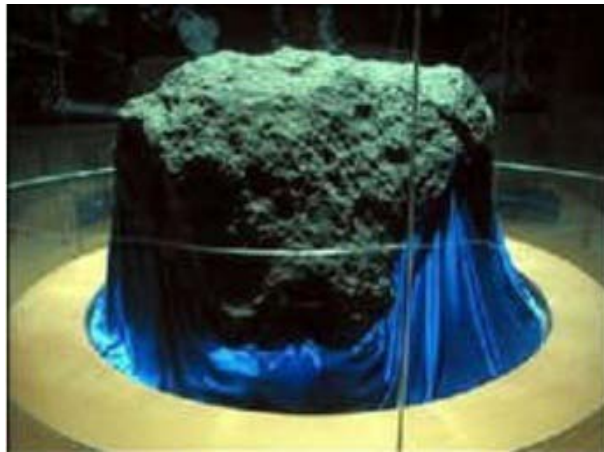
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◆ Meteorites in China

- **Jilin meteorites: 1976.3.8**, 3000 meteorites of total 2 ton mass have been recovered , one of the meteorites weighed as much as 1770kg.
- **Shangri-La, Yunnan: 2017.10.4** , many videos recorded.
- **Xishuangbanna, Yunnan: 2018.6.1** , many videos recorded.
- **Song Yuan, JiLin : 2019.10.11**, many videos recorded.



NO.1 meteor in Jilin meteorite event,
1976.3.8

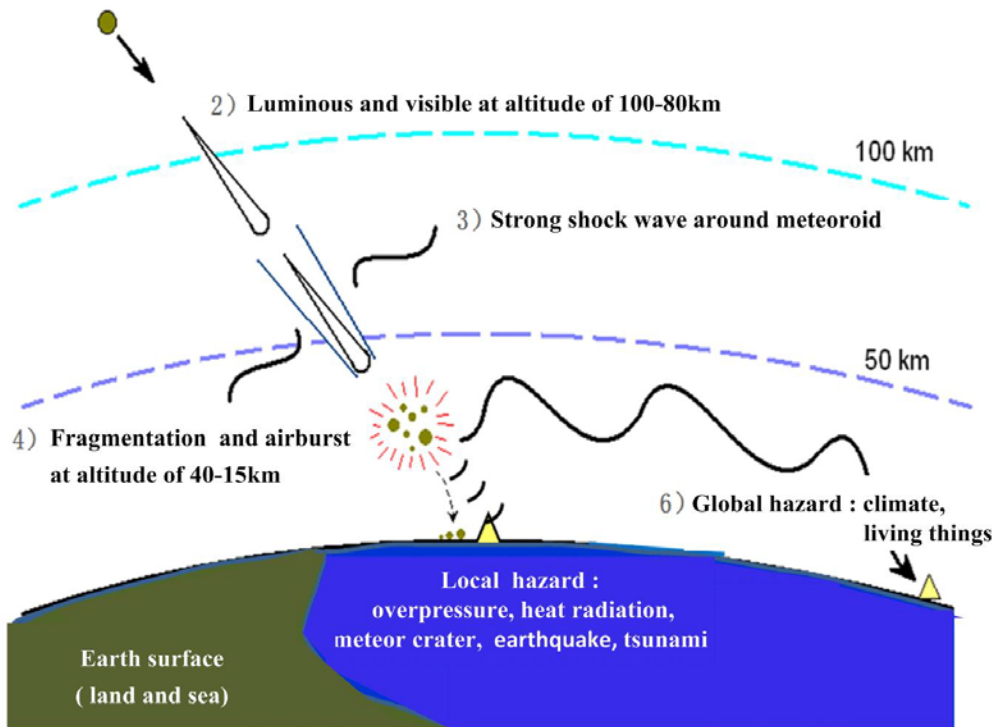


Shangri-La Yunnan, 2017.10.4

1) Earth impact by asteroids

➢ Velocity : ~20km/s(11.7-73km/s)

➢ Size : 0.1m-10km



- ◆ aerodynamic forces and trajectory during ultra-high velocity entry
- ◆ **aerodynamic heating during ultra-high velocity entry**
- ◆ ablation and thermal response of asteroid structure
- ◆ physical characteristics of asteroid entry process

Earth entry issues in planetary defense

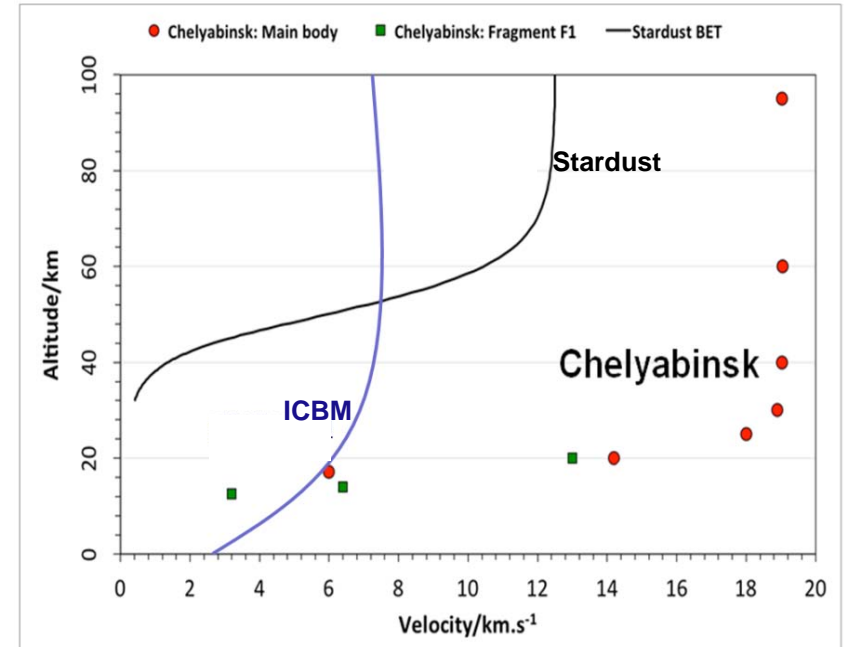


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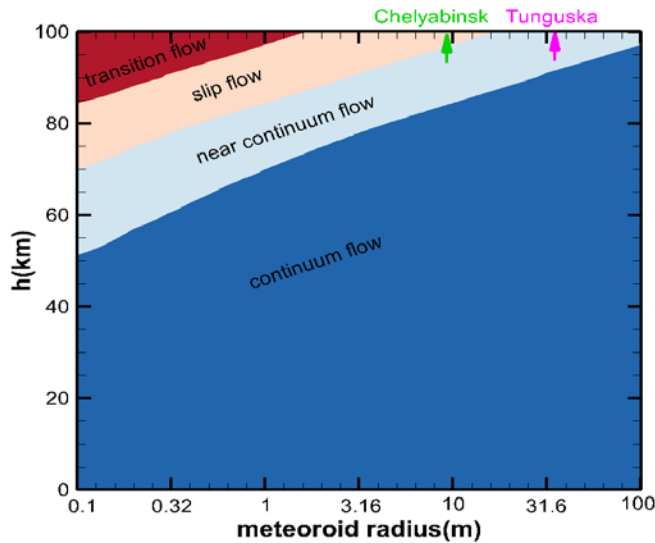
- ◆ Could the numerical methods and physical-chemical models purposed for reentry vehicles be used in asteroid entry problems?

| | Reentry vehicle | Asteroids |
|----------------------|-----------------------------|--|
| Size | <6m | up to several km |
| Shape | regular , smooth | irregular , unsmooth |
| Surface Materials | TPS Materials C/H/O/N/Si | Depending on the type (S, M, X) O/S/Si/Fe/Mg/Ca/Na/Al |
| Interior structure | few defect | cracks, cavities, voids |
| Entry speed (km/s) | 7.5~13 | 10~70 |
| Peak temperature | ≤20000K | ~50000K |

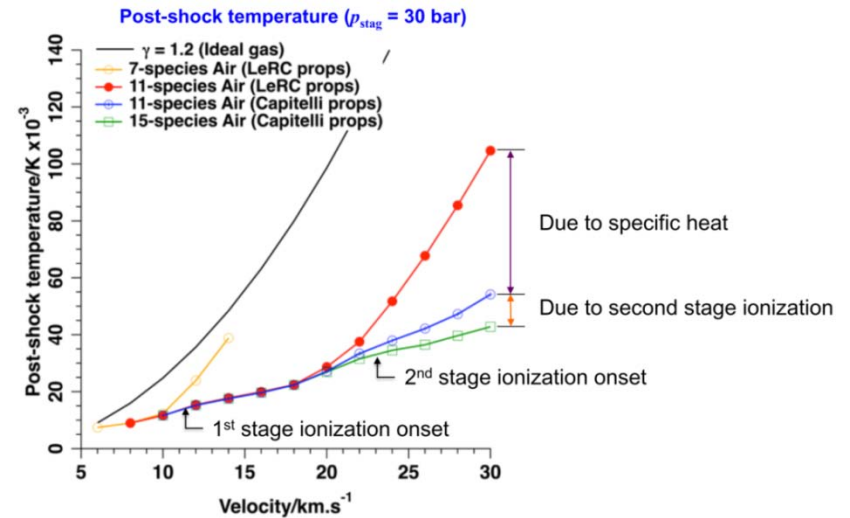


Comparison between the reentry vehicle and asteroid

- ◆ As the primary flow regime of asteroid's entry is continuum-near continuum, the CFD methods might be applicable for asteroid entry calculation;
- ◆ However, the physical-chemical models relevant to high temperature gas effects of traditional CFD should be confirmed and improved.



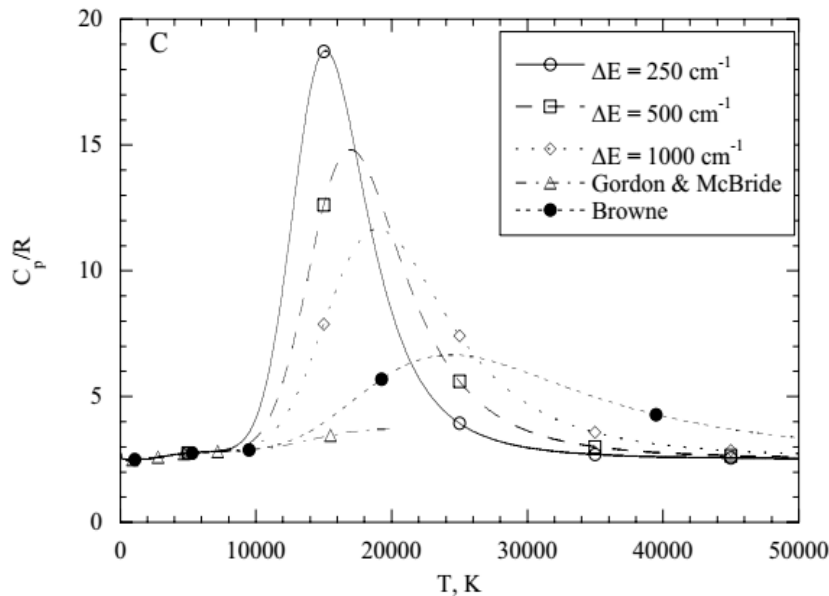
Flow regime of asteroid entry



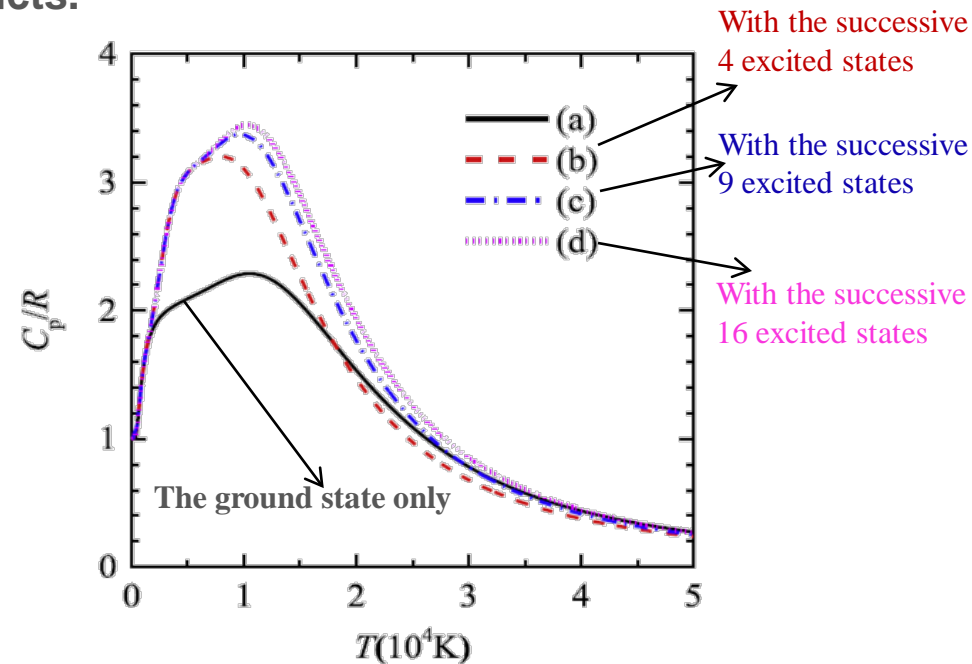
The effect of Air chemical model on post-shock temperature[1]

Thermodynamic properties of ultra-high-temperature gas

- ◆ Thermodynamics properties of ultra-high-temperature gas are sensitive to the number of electronic states, high-lying level energies and cut-off criteria.
- ◆ Thermodynamics properties database valid up to 50,000 K should be constructed for both air species and ablation products.



The influence of cut-off criteria on the C_p value of $C^{[2]}$



The influence of electronic states on the C_p value of $CN^{[3]}$

Ref 2: Capitelli.et.al. ESA STR246. Tables of internal partition functions and thermodynamic properties of high-temperature Mars-atmosphere species from 50K to 50000K

Ref 3: Z. Qin .et.al. JQSRT 210 (2018) 1–18 . High-temperature partition functions, specific heats and spectral radiative properties of diatomic molecules with an improved calculation of energy levels



Transport properties of ultra-high-temperature gas

- ◆ Based on the Chapman-Enskog method, the collision cross section are needed for transport properties computation.
- ◆ high accuracy database of collision cross section up to 50,000 K are needed for both air species and ablation products.

Table 2 Viscosity collision integral $\Omega^{2,2}$ (\AA^2) as a function of temperature for neutral-neutral and electron-neutral interactions in air

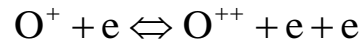
| Interaction | T, K | | | | | | | | | | | | Acc., % | Ref. |
|--------------------------------|-------|-------|-------|-------|------|------|------|------|------|--------|--------|--------|---------|--------|
| | 300 | 500 | 600 | 1000 | 2000 | 4000 | 5000 | 6000 | 8000 | 10,000 | 15,000 | 20,000 | | |
| N ₂ -N ₂ | 13.72 | — | 11.80 | 10.94 | 9.82 | 8.70 | — | 8.08 | 7.58 | 7.32 | — | — | 10 | 9 |
| N ₂ -O ₂ | 11.23 | — | — | 8.36 | 7.35 | 6.47 | 6.21 | — | — | 5.42 | 4.94 | — | 20 | 5 |
| N ₂ -NO | 13.44 | 11.87 | 11.44 | 10.48 | 9.32 | 8.04 | 7.61 | 7.27 | 6.74 | 6.33 | 5.62 | — | 25 | 24 |
| N ₂ -N | 11.21 | — | 9.68 | 8.81 | 7.76 | 6.73 | — | 6.18 | 5.74 | 5.36 | — | — | 10 | 8 |
| N ₂ -O | 8.99 | — | — | 6.72 | 5.91 | 5.22 | 5.01 | — | — | 4.36 | 3.95 | — | 20 | 5 |
| N ₂ -Ar | 11.97 | 10.50 | — | 9.28 | 8.38 | 7.53 | 7.25 | 7.02 | 6.63 | 6.32 | 5.73 | — | 20 | 5 |
| O ₂ -O ₂ | 12.62 | 11.06 | 10.65 | 9.72 | 8.70 | 7.70 | 7.38 | 7.12 | 6.73 | 6.42 | 5.89 | — | 20 | 24 |
| O ₂ -NO | 12.93 | 11.32 | 10.90 | 9.94 | 8.89 | 7.80 | 7.45 | 7.17 | 6.73 | 6.39 | 5.80 | — | 25 | 24 |
| O ₂ -N | — | 8.79 | 8.47 | 7.68 | 6.63 | 5.67 | 5.38 | 5.14 | 4.78 | 4.51 | 4.04 | — | 25 | 16 |
| O ₂ -O | 10.13 | — | 8.61 | 7.78 | 6.71 | 5.67 | — | 5.13 | 4.78 | 4.50 | — | — | 10 | 8 |
| O ₂ -Ar | 12.33 | 10.82 | — | 9.57 | 8.54 | 7.48 | 7.14 | 6.87 | 6.44 | 6.12 | 5.54 | — | 20 | 5 |
| NO-NO | 13.25 | 11.58 | 11.15 | 10.16 | 9.07 | 7.91 | 7.53 | 7.21 | 6.73 | 6.36 | 5.72 | — | 20 | 24 |
| NO-N | — | 9.65 | 9.26 | 8.29 | 7.07 | 5.94 | 5.60 | 5.33 | 4.91 | 4.60 | 4.06 | — | 25 | 16 |
| NO-O | — | 8.79 | 8.47 | 7.66 | 6.64 | 5.69 | 5.40 | 5.17 | 4.82 | 4.55 | 4.08 | — | 25 | 16 |
| NO-Ar | 13.03 | 11.33 | 10.89 | 9.90 | 8.85 | 7.79 | 7.44 | 7.15 | 6.71 | 6.37 | 5.77 | — | 25 | 24 |
| N-N | 9.11 | 7.94 | — | 6.72 | 5.82 | 4.98 | 4.70 | 4.48 | 4.14 | 3.88 | 3.43 | 3.11 | 5 | 15 |
| N-O | 9.08 | 8.15 | — | 7.09 | 6.06 | 5.14 | 4.88 | 4.67 | 4.34 | 4.07 | 3.56 | 3.21 | 5 | 15 |
| N-Ar | — | 9.31 | 8.96 | 8.03 | 6.84 | 5.75 | 5.42 | 5.17 | 4.77 | 4.47 | 3.95 | — | 20 | 16 |
| O-O | 9.46 | 8.22 | — | 6.76 | 5.58 | 4.67 | 4.41 | 4.20 | 3.88 | 3.64 | 3.21 | 2.91 | 5 | 15 |
| O-Ar | 11.11 | 10.13 | — | 8.87 | 7.69 | 6.60 | 6.26 | 6.00 | 5.59 | 5.28 | 4.74 | 4.37 | 20 | 5 |
| Ar-Ar | 12.88 | 11.12 | — | 9.66 | 8.60 | 7.57 | 7.23 | 6.96 | 6.53 | 6.20 | 5.60 | 5.17 | 5 | 24 |
| e-N ₂ | — | 1.46 | — | 2.07 | 2.96 | 3.88 | 4.09 | 4.15 | 4.04 | 3.85 | 3.41 | 3.12 | 25 | 28 |
| e-O ₂ | — | — | — | 1.30 | 1.73 | 2.10 | 2.18 | 2.23 | 2.29 | 2.31 | 2.32 | 2.31 | 20 | 28 |
| e-NO | — | — | — | — | 5.64 | 4.52 | 4.05 | 3.73 | 3.37 | 3.18 | 2.92 | 2.75 | 35 | 28 |
| e-N | — | — | — | — | 5.68 | 3.71 | 3.52 | 3.42 | 3.30 | 3.20 | 2.95 | 2.58 | 35 | 39, 40 |
| e-O | — | — | — | 0.82 | 1.05 | 1.34 | 1.44 | 1.52 | 1.65 | 1.73 | 1.85 | 1.90 | 30 | 27 |
| e-Ar | — | — | — | 0.17 | 0.30 | 0.79 | 1.05 | 1.31 | 1.81 | 2.25 | 3.07 | 3.48 | 15 | 26 |

The estimated accuracy of collision cross section data [4]

Chemical kinetics of ultra-high-temperature gas

- ◆ Chemical kinetics models for vehicle entry should not be used directly for asteroid entry.
- ◆ New chemical kinetics such as multiple stage ionization、photochemical kinetics and chemical kinetics for asteroid ablation products should be developed.

➤ Chemical kinetics for multiple stage ionization



➤ Photochemical kinetics

➤ Chemical kinetics for ablation products

[5]

Table 1: Photochemical processes applied in the present study.

| # | Process | Spectral Range | Data Source |
|---|--|----------------|-----------------------------------|
| 1 | N ₂ Photodissociation: N ₂ + hν ↔ 2N | 9.8 eV < hν | Stanley and Carlson ³⁰ |
| 2 | O ₂ Photodissociation: O ₂ + hν ↔ 2O | 7.1 eV < hν | Mnatsakanyan ³¹ |
| 3 | N ₂ Photoionization: N ₂ + hν ↔ N ₂ ⁺ + e ⁻ | 12.4 eV < hν | Romanov et al. ³² |
| 4 | O ₂ Photoionization: O ₂ + hν ↔ O ₂ ⁺ + e ⁻ | 9.7 eV < hν | Romanov et al. ³² |
| 5 | N Photoionization: N + hν ↔ N ⁺ + e ⁻ | 12.4 eV < hν | TOPbase ³³ |
| 6 | O Photoionization: O + hν ↔ O ⁺ + e ⁻ | 9.7 eV < hν | TOPbase ³³ |

Table 2: Chemical kinetics for meteor ablation products.

| \bar{i} | Reaction | $A_{f,i,t}$ | $n_{f,i,t}$ | $D_{f,i,t}$ | Ref. |
|-----------|---|-------------|-------------|-------------|-----------------------------------|
| 1 | Si + e ⁻ ↔ Si ⁺ + e ⁻ + e ⁻ | 2.5e+34 | -3.82 | 9.46e+4 | Based on N rate ¹⁷ |
| 2 | Fe + e ⁻ ↔ Fe ⁺ + e ⁻ + e ⁻ | 2.5e+34 | -3.82 | 9.17e+4 | Based on N rate ¹⁷ |
| 3 | Mg + e ⁻ ↔ Mg ⁺ + e ⁻ + e ⁻ | 2.5e+34 | -3.82 | 8.87e+4 | Based on N rate ¹⁷ |
| 4 | Si + NO ↔ SiO + N | 3.2e+13 | 0.0 | 1775.0 | Mick et al. ³⁸ |
| 5 | Si + O ₂ ↔ SiO + O | 2.1e+15 | -0.53 | 16.83 | Le Picard et al. ³⁹ |
| 6 | SiO + M ↔ Si + O + M | 4.0e+14 | 0.0 | 9.56e+4 | Estimate |
| 7 | SiO ₂ + M ↔ SiO + O + M | 4.0e+14 | 0.0 | 9.56e+4 | Estimate |
| 8 | Fe + O ₂ ↔ FeO + O | 1.3e+14 | 0.0 | 1.02e+4 | Akhmadov et al. ⁴⁰ |
| 9 | Mg + O ₂ ↔ MgO + O | 5.1e+10 | 0.0 | 0.0 | Hodgson and Mackie ⁴¹ |
| 10 | SO + O ↔ S + O ₂ | 2.4e+07 | 1.51 | 2.53e+3 | Lu et al. ⁴² |
| 11 | SO ₂ + S ↔ SO + SO | 4.8e+14 | 0.0 | 1.08e+4 | Murakami ⁴³ |
| 12 | O ₂ + SO ↔ SO ₂ + O | 2.3e+12 | 0.0 | 3.70e+3 | Garland ⁴⁴ |
| 13 | Al + O ₂ ↔ AlO + O | 2.0e+13 | 0.0 | 0.0 | Cohen and Westberg ⁴⁵ |
| 14 | Al + SO ₂ ↔ SO + AlO | 9.6e+13 | 0.0 | 2.00e+3 | Fontijn and Felder ⁴⁵ |
| 15 | Al + e ⁻ ↔ Al ⁺ + e ⁻ + e ⁻ | 2.5e+19 | -0.82 | 6.94e+4 | Based on N rate ¹⁷ |
| 16 | NaO + O ↔ Na + O ₂ | 2.2e+14 | 0.0 | 0.0 | Plane and Husain ⁴⁷ |
| 17 | Na + e ⁻ ↔ Na ⁺ + e ⁻ + e ⁻ | 2.5e+19 | -0.82 | 5.96e+4 | Based on N rate ¹⁷ |
| 18 | Ca + O ₂ ↔ CaO + O | 2.5e+14 | 0 | 7.25e+3 | Kashireninov et al. ⁴⁸ |
| 17 | Ca + e ⁻ ↔ Ca ⁺ + e ⁻ + e ⁻ | 2.5e+19 | -0.82 | 7.09e+4 | Based on N rate ¹⁷ |

Internal degree relaxation model of ultra-high-temperature gas

- ◆ Park's two-temperature model ($T_{tr}-T_{ve}$), Multi-temperature model ($T_t-T_r-T_{vi}-T_{el}$), State-To-State model.
- ◆ model parameter such as vibrational relaxation time need to be validated at ultra-high-temperature.

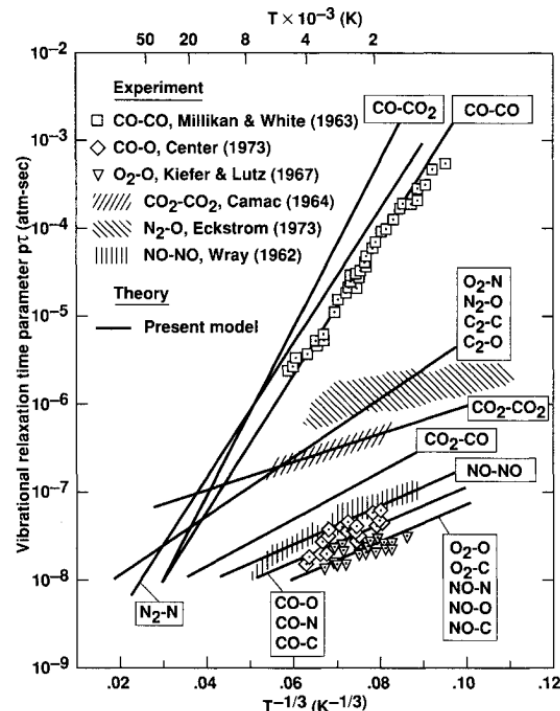


Fig. 2 Comparison between the measured vibrational relaxation times^{7,8,51-53} and the present model for carbon-containing species. [6]

Radiation model of ultra-high-temperature gas

- ◆ Radiation computation: line-by-line method, e.g. NEQAIR.
- ◆ Radiation database: TOPbase 、 HITEMP. Radiation band of asteroid ablation products should be developed.

Table 3: Summary of molecular band modeling for meteor ablation products[5]

| Specie | Transition | Spectral Range (eV) | Data Source |
|--------|------------|---------------------|--|
| SiO | A-X | 4.54-5.79 | Franck-Condon factors and energy levels from Geier et al., ⁵¹ and band oscillator strength from Park and Arnold. ⁵² |
| SiO | E-X | 5.74-7.55 | Franck-Condon factors and band oscillator strength taken from Naidu et al. ⁵³ and Drira, ⁵⁴ and energy levels from Lagerqvist. ⁵⁵ |
| FeO | Orange | 1.68-2.38 | Oscillator strengths and energy levels taken from Michels. ⁵⁶ |
| MgO | B-A | 1.72-2.45 | Oscillator strengths and energy levels taken from Daily ⁵⁷ and Bell et al. ⁵⁸ |
| MgO | D-A | 1.72-2.45 | Oscillator strengths and energy levels taken from Naulin et al. ⁵⁹ and Bell et al. ⁵⁸ |
| MgO | B-X | 2.38-2.69 | Oscillator strengths and energy levels taken from Daily ⁵⁷ and Bell et al. ⁵⁸ |
| CaO | A-X | 1.1-2.0 | Oscillator strengths and energy levels taken from Doherty ⁶⁰ and Liszt. ⁶¹ |
| CaO | B-X | 2.6-3.7 | Oscillator strengths and energy levels taken from Pasternack ⁶² and Liszt. ⁶¹ |
| CaO | Orange | 1.7-2.2 | Oscillator strengths and energy levels taken from Pasternack ⁶² and Liszt. ⁶¹ |
| CaO | Green | 1.7-2.2 | Oscillator strengths and energy levels taken from Pasternack, ⁶² Liszt, ⁶¹ and Baldwin. ⁶³ |
| SO | A-X | 3.8-5.0 | Franck-Condon factors and band oscillator strength taken from Borin ⁶⁴ and energy levels from Rosen. ⁶⁵ |
| AlO | B-X | 2.2-3.0 | Oscillator strengths and energy levels taken from Borovicka. ⁶⁶ |



□ AHENS

- The CFD code AHENS , developed at our institute for aerothermal environment simulation, solves the NS equations on structured grids.

□ Numerical Method

- The finite-volume method is used to discretize the governing equations.
- Hybrid Steger-Warming and Godunov flux scheme with second-order spatial accuracy.
- LU-SGS implicit algorithm for time step iteration.
- MPI based parallelism.

□ High-temperature Gas Model

- 1T / 2T /3T Multi-species gas mixture model.
- Finite-rate chemistry model.
- Polynomial fitting method for thermodynamic properties.
- Gupta's model with collision cross section data for transport coefficients.
- Modified Fick's model for mass diffusion.

$$J_{s \neq e} = -\rho D_s \nabla Y_s - Y_s \sum_{r \neq e} -\rho D_r \nabla Y_r \quad J_e = -\frac{1}{q_e} \sum_{s \neq e} q_s J_s$$

It ensures that the sum of mass fluxes is zero.



□ Boundary condition

- Freestream inflow, extrapolation outflow, symmetry.
- RCS jet boundary condition, stagnation boundary condition.
- Unslip or slip boundary condition.
- Isothermal wall or radiative equilibrium wall condition.
- Quasi steady ablation with finite-rate surface chemistry .
- Catalytic wall condition: non-catalytic , super-catalytic, specified catalytic coefficients.

□ Radiation model

- Non-Boltzmann models for diatomic molecules and atomic species electronic state populations
- Line information and cross-sections following the work of Johnston.^[7]

| Molec. | Transition | Name | Spectral Range (eV) |
|-----------------------------|--|--|---------------------|
| N ₂ | B ³ Π _g - A ³ Σ _u ⁺ | 1 ⁺ (1 st -positive) | 0.2 - 2.5 |
| N ₂ | C ³ Π _u - B ³ Π _g | 2 ⁺ (2 nd -positive) | 2.7 - 4.7 |
| N ₂ | c ₄ ³ Σ _u ⁺ - X ¹ Σ _g ⁺ | Carroll-Yoshino | 11.5 - 14.0 |
| N ₂ | c ₃ ³ Π _u - X ¹ Σ _g ⁺ | Worley-Jenkins | 11.5 - 14.0 |
| N ₂ | b ¹ Π _u - X ¹ Σ _g ⁺ | Birge-Hopfield I | 7.0 - 13.1 |
| N ₂ | b ¹ Σ _u ⁺ - X ¹ Σ _g ⁺ | Birge-Hopfield II | 7.6 - 14.0 |
| N ₂ | o ₃ ¹ Π _u - X ¹ Σ _g ⁺ | Worley | 10.4 - 14.0 |
| N ₂ ⁺ | B ² Σ _u ⁺ - X ² Σ _g ⁺ | 1 ⁻ (1 st -negative) | 1.2 - 4.6 |
| NO | B ² Π _r - X ² Π _r | β (beta) | 2.1 - 6.9 |
| NO | A ² Σ ⁺ - X ² Π _r | γ (gamma) | 3.2 - 7.5 |
| NO | C ² Π _r - X ² Π _r | δ (delta) | 3.7 - 7.6 |
| NO | D ² Σ ⁺ - X ² Π _r | ε (epsilon) | 3.4 - 8.0 |
| NO | B ² Δ - X ² Π _r | β' (beta-prime) | 3.9 - 8.4 |
| NO | E ² Σ ⁺ - X ² Π _r | γ' (gamma-prime) | 4.6 - 8.9 |
| O ₂ | B ³ Σ _u ⁻ - X ³ Σ _g ⁻ | Schumann-Runge | 2.6 - 7.0 |

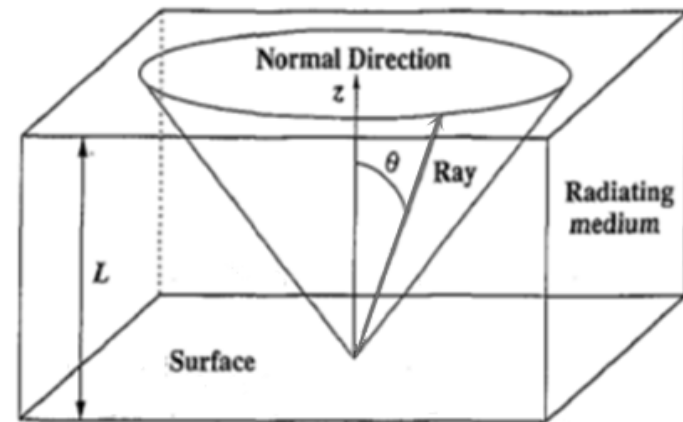
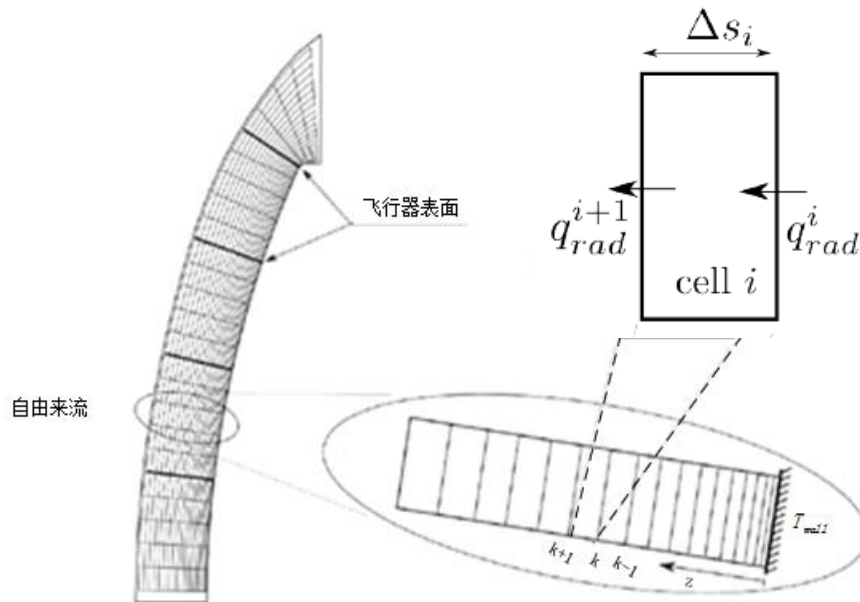
molecular band systems

□ Radiation Transfer

- Radiative Transfer Equation:

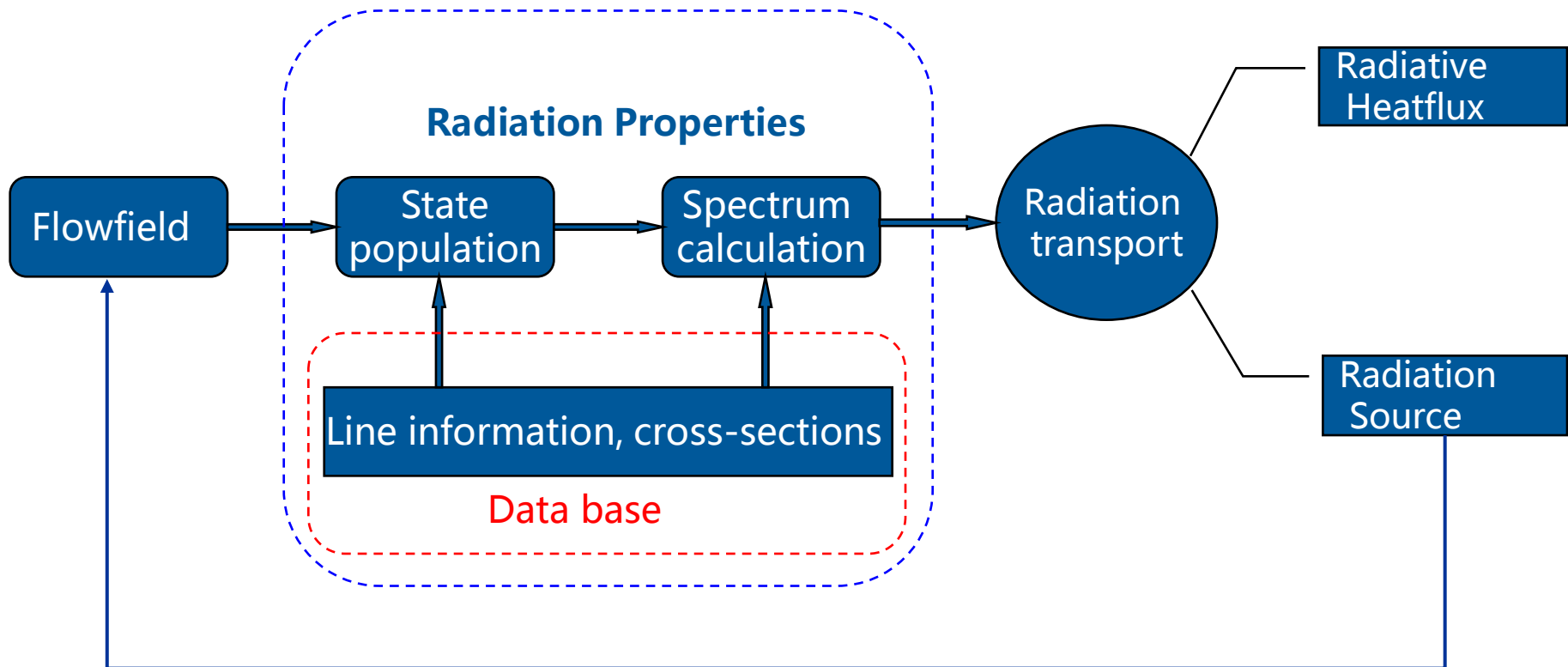
$$\frac{dI_\nu(s, \Omega)}{ds} = j_\nu(s) - \kappa'_\nu(s)I_\nu(s, \Omega)$$

- Tangent Slab approximation:



$$-(\nabla \cdot \vec{q}_{rad})_i = -\left(\frac{\partial q_{rad}}{\partial s}\right)_i \approx \frac{-(q_{rad}^{(i+1)} - q_{rad}^{(i)})}{\Delta s_i}$$

Loosely coupled approach



Coupled simulation of flow and radiation



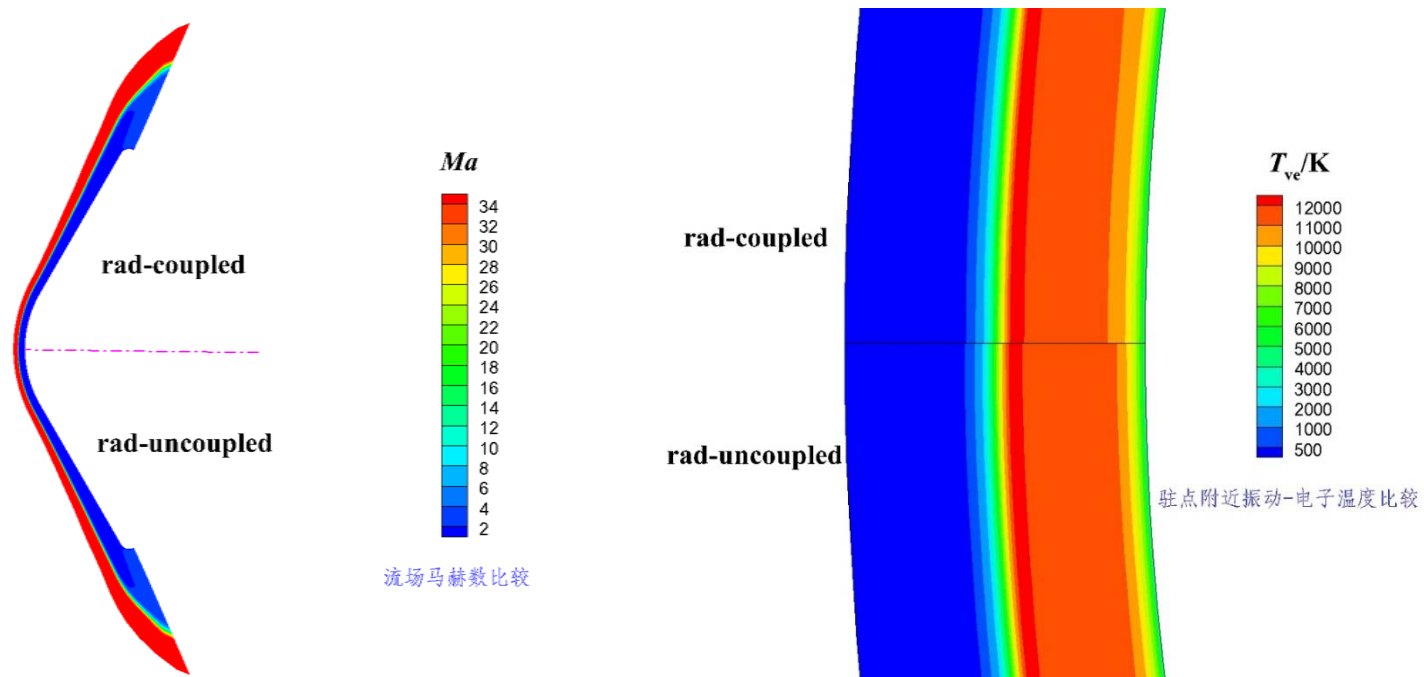
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□ Stardust vehicle reentry

- ◆ The radiation-flow coupling effect is relatively small to the flowfield of the reentry vehicle.

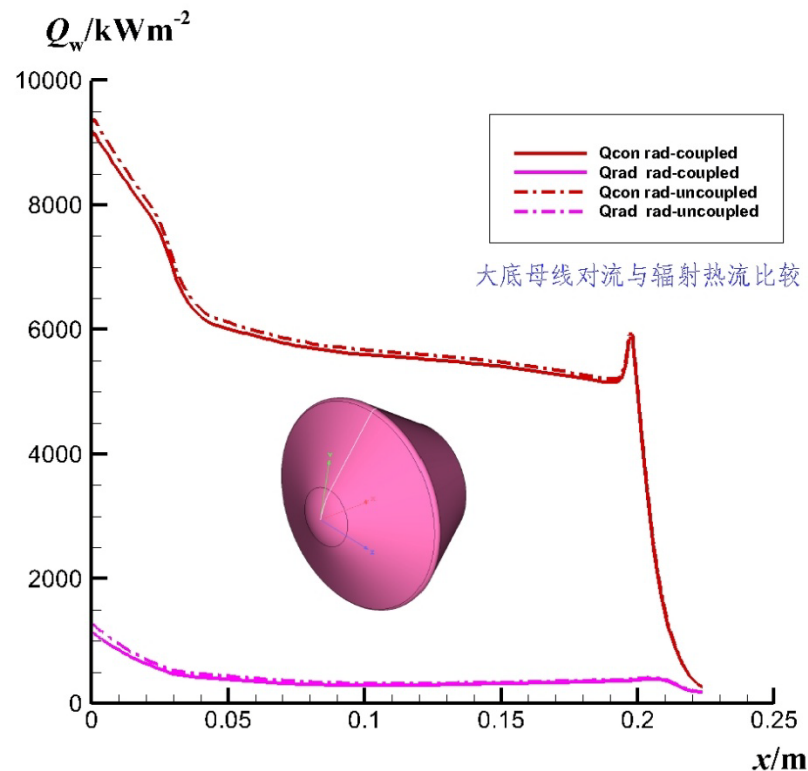
| case | H(km) | V _{oo} (m/s) | AoA(°) | Wall conditon |
|------|-------|-----------------------|---------|-----------------|
| 1 | 59.77 | 11136.7 | 0 | super-catalytic |



Comparison of the predicted flow properties

□ Stardust Capsule reentry

- ◆ The convective is the dominated mechanism of aerodynamic heating for reentry vehicles, un-coupled radiation simulation can be adopted.



Comparison of the predicted heat flux

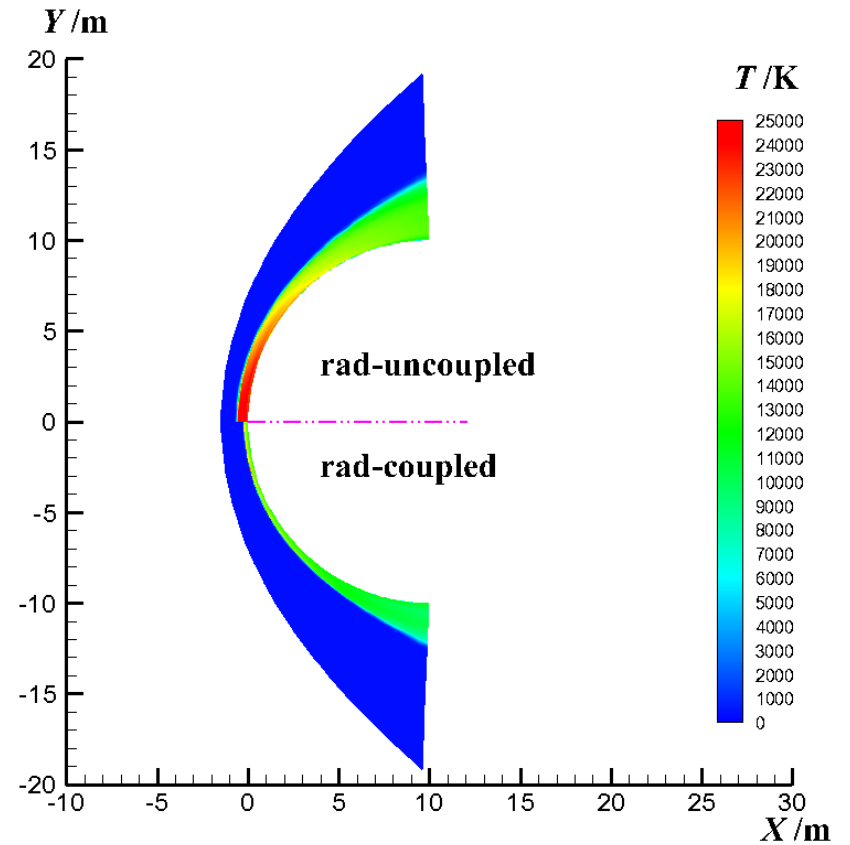
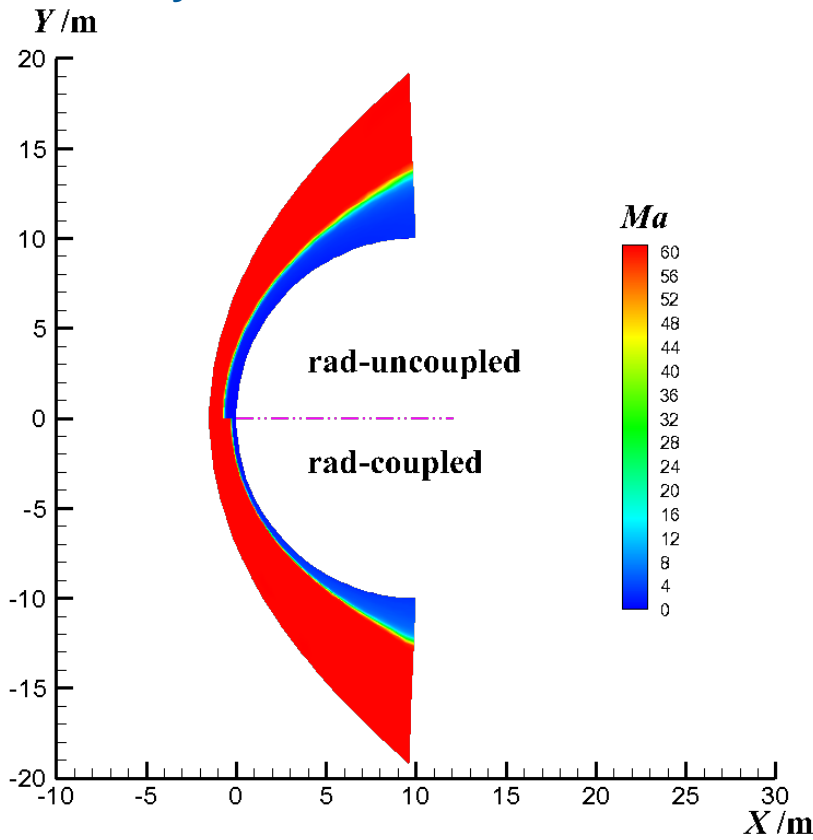
□ Asteroid entry

- Coupled and uncoupled radiation simulations are carried out for typical asteroid entry condition to investigate the influence mechanism radiation-flow coupling on aerodynamic heating.
- A 13 species ($N_2, O_2, N, O, NO, NO^+, N_2^+, N^+, N^{++}, O_2^+, O^+, O^{++}, e^-$) ionized air model is incorporated. [8]

| case | H(km) | V ₀₀ (km/s) | Wall conditon | D(m) |
|------|-------|------------------------|-----------------|------|
| 1 | 50 | 15、 20 | super-catalytic | 20 |
| 2 | 50 | 15、 20 | super-catalytic | 50 |
| 3 | 50 | 15、 20 | super-catalytic | 100 |
| 4 | 50 | 15、 20 | super-catalytic | 140 |

□ Asteroid entry

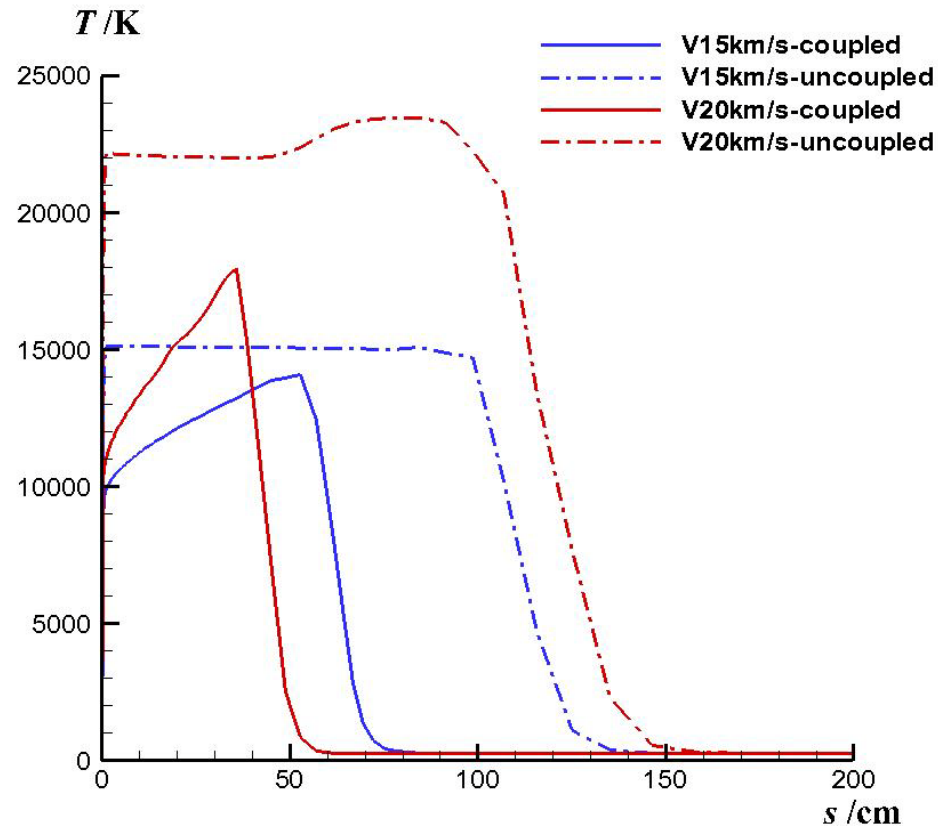
- ◆ Radiation-flow coupling effect plays a significant role in flow structure of asteroid entry.
- ◆ Compared with uncoupled case, the temperature and thickness of shock layer are much smaller.



D=20m, V=20km/s

□ Asteroid entry

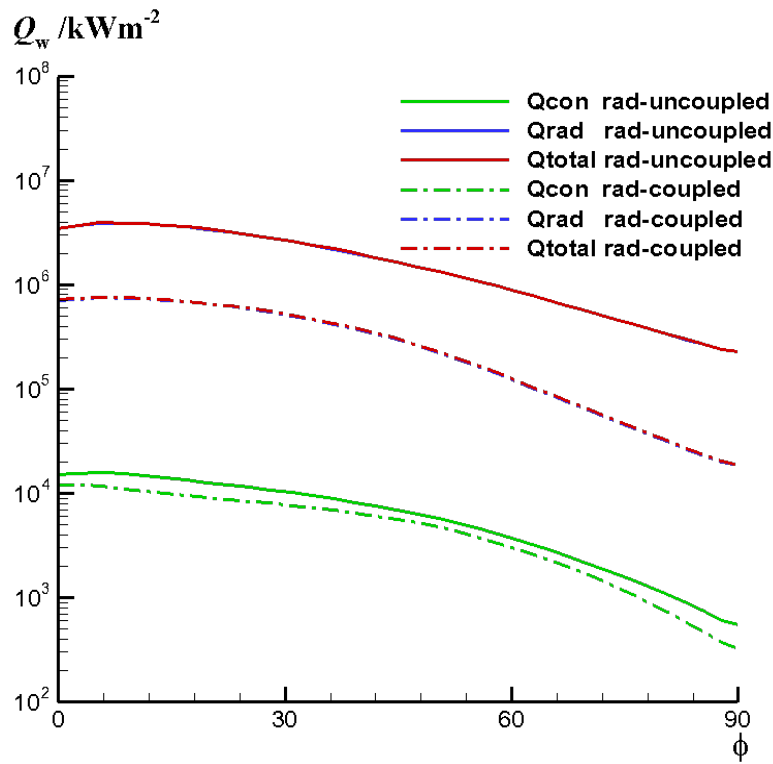
- ◆ With the increase of entry velocity, the coupling effects on flowfield is enhanced, and the variations of shock standoff distance and peak temperature are enlarged.



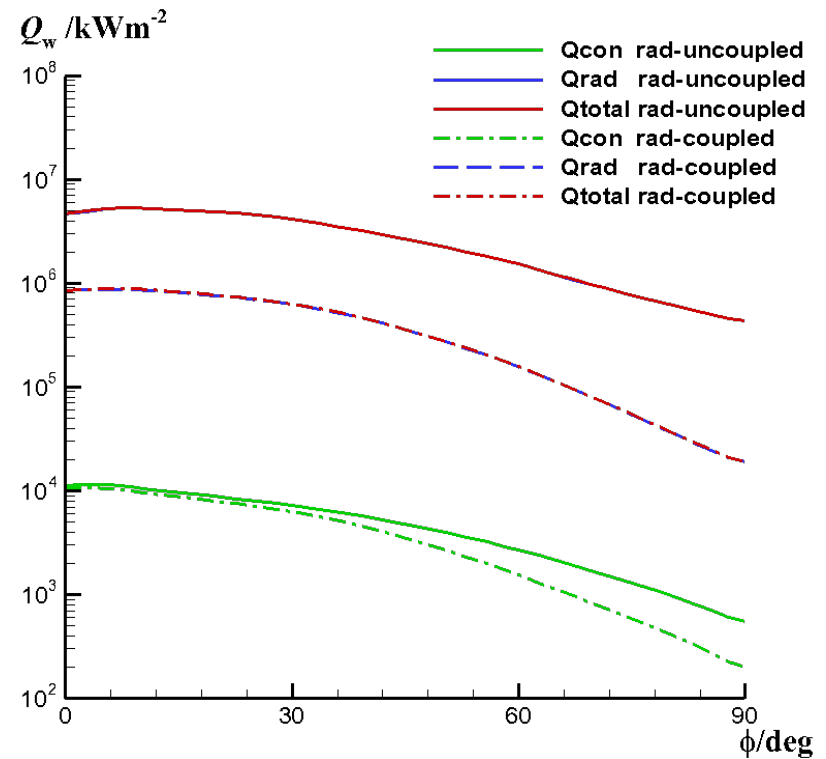
Flow temperature along stagnation line(D=50m)

□ Asteroid entry

◆ The radiative cooling and thin shock layer are the physical mechanism that radiation-flow coupling will ease aero-heating.



D=20m, V=20km/s

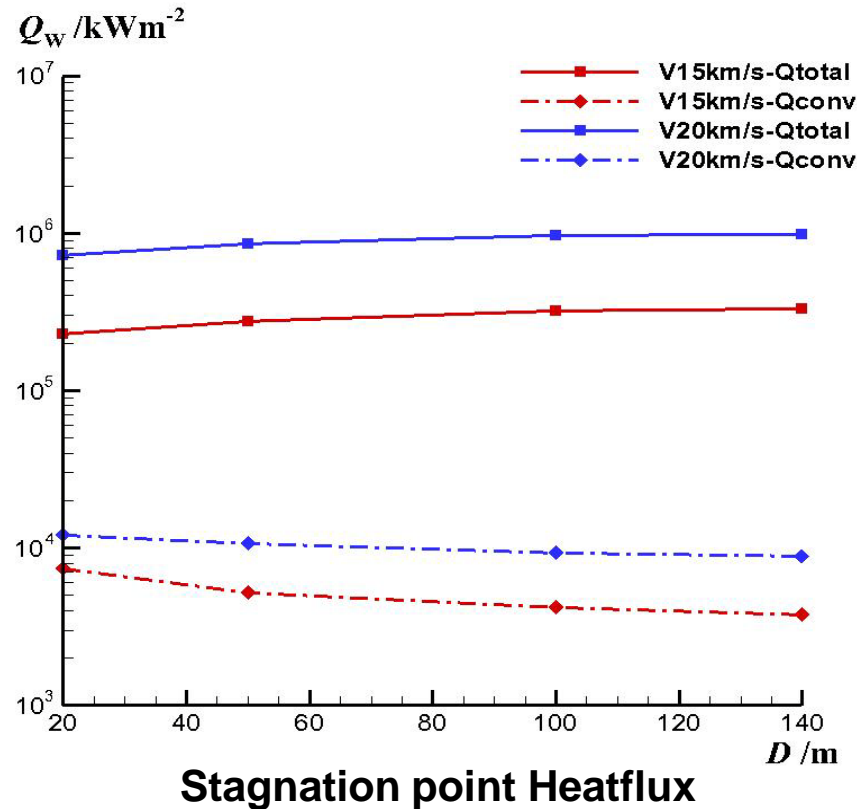


D50m, V=20km/s

Heatflux along surface

□ Asteroid entry

- ◆ There are obvious differences of heating character between asteroid entry and hypersonic vehicle reentry:
 - The radiative heating is dominant for asteroid entry and the convective heating is negligible.
 - Total heat flux at stagnation point increases with diameter.



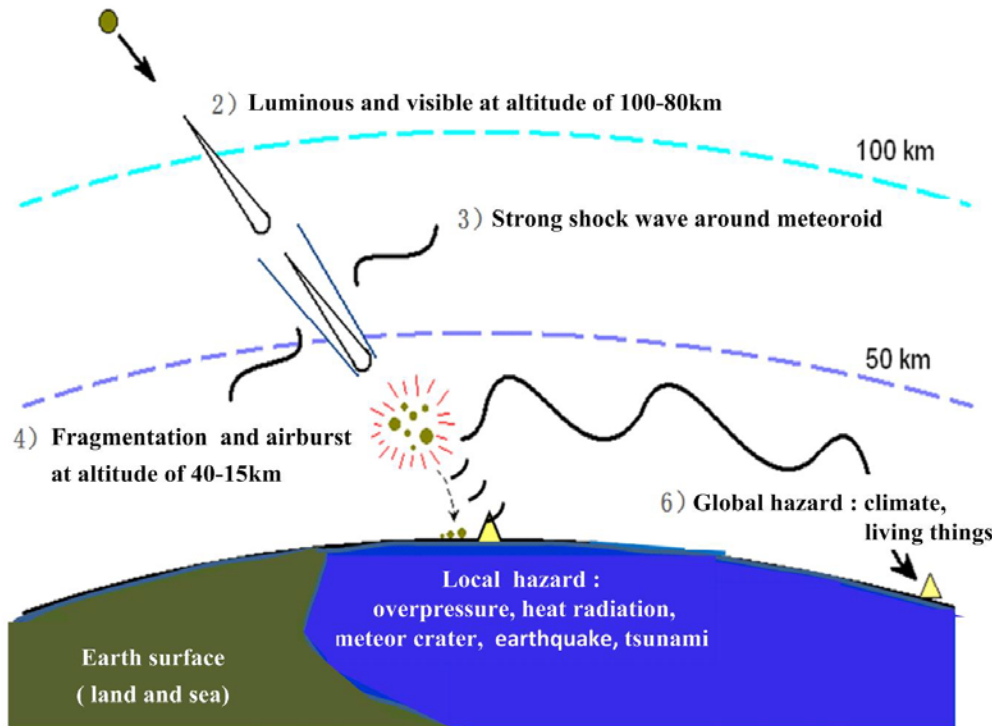


Outline

- 1、 Background
- 2、 Numerical method and physical-chemical model
- 3、 Results and discussion
- 4、 Summary**

(1) Hypervelocity aerothermodynamics plays an important role in the analysis of asteroid impacting the Earth.

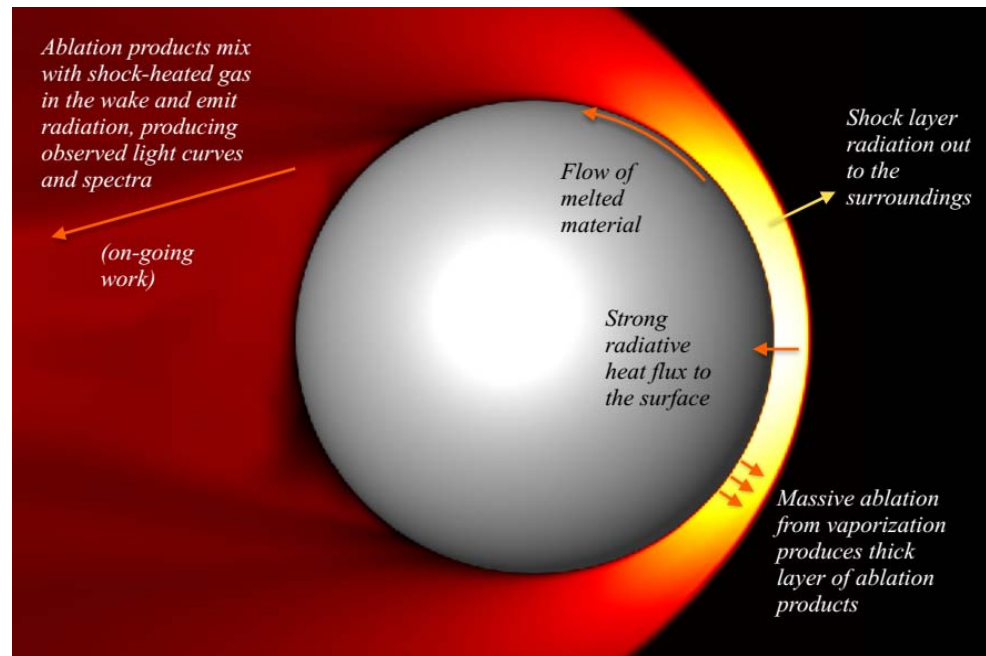
- 1) Earth impact by asteroids
 - Velocity : ~20km/s(11.7-73km/s)
 - Size : 0.1m-10km



- ◆ aerodynamic forces and trajectory during ultra-high velocity entry
- ◆ aerodynamic heating during ultra-high velocity entry
- ◆ ablation and thermal response of asteroid structure
- ◆ physical characteristics of asteroid entry process

(2) In order to investigate the aerothermodynamic problems in asteroid entry to Earth, the high temperature gas models, such as thermo-chemical models and radiation data, need to be extended to greater than 20000K.

(3) Since there are strong coupling effects among flow-field, radiation and ablation in Earth entry problem of asteroid, both the coupling mechanism and numerical methods are needed to be further explored.



The coupling between asteroid entry flow, radiation and ablation [7]



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Thanks for your attention

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and Development Center